

# STIC Database Tracking Number: 103062

TO: Molly Ceperley

Location: CM1/8D15&7E12

**Art Unit: 1641** 

September 9, 2003

Case Serial Number: 09/099048

From: P. Sheppard Location: CM1-1E03 Phone: (703) 308-4499

sheppard@uspto.gov

Search Notes	
	·
	•

(Acops DB# 103062

# SEARCH REOUEST FORM

Scientific and Technical Information Center

(3)73)
Requester's Full Name: Molly CEPERLEY Examiner #: 59757 Date: 09/04/03
Art Unit: 1641 Phone Number 308-4239 Serial Number: Reissue 091099 048
Mail Box and Bldg/Room Location: 8D15 Results Format Preferred (circle) PAPER DISK E-MAIL
ME!
If more than one search is submitted, please prioritize searches in order of need.
Please provide a detailed statement of the search topic, and describe as specifically as possible the subject matter to be searched. Include the elected species or structures, keywords, synonyms, acronyms, and registry numbers, and combine with the concept or utility of the invention. Define any terms that may have a special meaning. Give examples or relevant citations, authors, etc, if known. Please attach a copy of the cover sheet, pertinent claims, and abstract.
Title of Invention: Rate Measurements of Biomolecular Reactions Using Electrochemilum escence
Inventors (please provide full names): Laurette Macamulli, Jonathan K. heland
Stephanie A. Hayes (Igen Tinc.)
Earliest Priority Filing Date: 12/02/94
*For Sequence Searches Only* Please include all pertinent information (parent, child, divisional, or issued patent numbers) along with the appropriate serial number.
Please search for:
an electrochemiluminescent (ECL) method for determining the time course of a reaction (see the attached claims).
@ Electrochemiluminescence (also luminophore)
Time course of reaction (or reaction rate, rate of reaction, time interval) pre-selected, voltage pulses, modulated signal, demodulated signal, enzymatic reaction, time course, voltage pulses, are measurement)
Approved

STAFF USE ONLY	Type of Search	Vendors and cost where applicable								
Searcher Skymank	NA Sequence (#)	STN								
Searcher Phone #: 38-4499	` AA Sequence (#)	Dialog								
Searcher Location:	Structure (#)	Questel/Orbit								
Date Searcher Picked Up:	Bibliographic	Dr.Link								
Date Completed: 4/9/03	Litigation	Lexis/Nexis								
Searcher Prep & Review Time:	Fulltext	Sequence Systems								
Clerical Prep Time:	Patent Family	WWW/Internet								
Online Time:	Other	Other (specify)								

BEST AVAILABLE COPY

=> fil hcaplus FILE 'HCAPLUS' ENTERED AT 17:29:09 ON 08 SEP 2003 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2003 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

s e considered of ching cring

FILE COVERS 1907 - 8 Sep 2003 VOL 139 ISS 11 FILE LAST UPDATED: 7 Sep 2003 (20030907/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

```
=>
=> d stat que
               1 SEA FILE=REGISTRY ABB=ON PLU=ON ELECTROCHEM/BI
13 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINOPHOR/BI
L1
L2
                1 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINOPHORE/BI
L3
           13142 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINESC?
T.7
          275185 SEA FILE=HCAPLUS ABB=ON PLU=ON L1 OR ?ELECTROCHEM?
          230077 SEA FILE=HCAPLUS ABB=ON PLU=ON L7 OR ?LUMINESC? 4251 SEA FILE=HCAPLUS ABB=ON PLU=ON L8 (L) L9
L9
L10
L11
                  SEL
                       PLU=ON L3 1- CHEM:
                                                       2 TERMS
                1 SEA FILE=HCAPLUS ABB=ON PLU=ON L11
L12
L13
            3990 SEA FILE=HCAPLUS ABB=ON
                                              PLU=ON L12 OR L2 OR ?LUMINOPHOR?
               48 SEA FILE=HCAPLUS ABB=ON PLU=ON L10 AND L13
L14
               31 SEA FILE=HCAPLUS ABB=ON PLU=ON L14 AND (REACTION OR COURSE
T_115
                  OR BINDING)
```

=> =>

=>

=> d ibib abs hitrn 115 1-31

L15 ANSWER 1 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2002:691907 HCAPLUS

DOCUMENT NUMBER: 138:286894

TITLE: Electrochemical light, from laboratory curiosity to

useful analytical technique

AUTHOR(S): Richter, Mark M.

CORPORATE SOURCE: Dept. of Chemistry, Southwest Missouri State Univ.,

Springfield, MO, 65804, USA

SOURCE: Chemical Educator [online computer file] (2002), 7(4),

195-199

CODEN: CHEDF5; ISSN: 1430-4171

URL: http://link.springer-

ny.com/link/service/journals/00897/papers/0007004/7401

95mr.pdf

PUBLISHER: Springer-Verlag New York Inc.

Journal; (online computer file) DOCUMENT TYPE: English LANGUAGE: Electrochemiluminescence (ECL) is the process by which species generated at electrodes undergo electron-transfer reactions to form excited states that emit light. The application of a voltage to an electrode in the presence of an ECL luminophore, such as Ru(bpy)32+ (where bpy = 2,2'-bipyridine) or diphenylanthracene, results in light emission and allows detection of the emitter at very low concns. (.ltoreq. 10-12 mol dm-3). By employing ECL-active species as labels on biol. mols., ECL has found com. application for immunoassays and DNA analyses. The history of ECL is presented including the earliest, curiosity driven expts. and the development of ECL into an anal. technique for clin. diagnostic applications. The development and use of ECL sensors is an excellent example of how, over time, a lab. curiosity can become a useful, powerful, and com. viable technique. REFERENCE COUNT: 80 THERE ARE 80 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L15 ANSWER (2) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN 2002:571839 HCAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: 138:234169 TITLE: Electrochemiluminescence AUTHOR(S): Richter, Mark M. Department of Chemistry, Southwest Missouri State University, Springfield, MO, 65804-0089, USA CORPORATE SOURCE: Optical Biosensors (2002), 173-205. Editor(s): SOURCE: Ligler, Frances S.; Rowe Taitt, Chris A. Elsevier Science B.V.: Amsterdam, Neth. CODEN: 69CXMQ; ISBN: 0-444-50974-7 DOCUMENT TYPE: Conference; General Review LANGUAGE: English A review. Electrochemiluminescence (ECL) is the process where AB species generated at electrodes undergo electron transfer reactions to form excited states that emit light. Application of a voltage to an electrode in the presence of an ECL luminophore such as Ru(bpy)32+ (where bpy = 2,2'-bipyridine) results in light emission and allows detection of the emitter at very low concns. (<10-11 M). employing ECL-active species as labels on biol. mols., ECL has found application in com. instruments for immunoassays and DNA analyses. Com. systems have been developed that use ECL to detect many clin. relevant analytes with high sensitivity and selectivity. The principles, history, applications, advantages, limitations and possibilities for improving the performance of this technol. are outlined in this chapter. REFERENCE COUNT: THERE ARE 164 CITED REFERENCES AVAILABLE FOR 164 THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L15 ANSWER (3) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 2002:98150 HCAPLUS DOCUMENT NUMBER: 136:291175 TITLE: Time-resolved electrochemiluminescence of platinum(II) coproporphyrin AUTHOR(S): Canty, P.; Vare, L.; Hakansson, M.; Spehar, A.-M.; Papkovsky, D.; Ala-Kleme, T.; Kankare, J.; Kulmala, S. Laboratory of Inorganic and Analytical Chemistry, CORPORATE SOURCE: Helsinki University of Technology, FIN-02015, Finland Analytica Chimica Acta (2002), 453(2), 269-279 SOURCE: CODEN: ACACAM; ISSN: 0003-2670 PUBLISHER: Elsevier Science B.V. DOCUMENT TYPE: Journal LANGUAGE: English

Cathodic pulse polarization of oxide-covered aluminum electrodes can generate **electrochemiluminescence** (ECL) from metalloporphyrins.

This is based on the tunnel emission of hot electrons into aq. electrolyte soln., which probably results in the generation of hydrated electrons as reducing mediators. These tunnel emitted electrons allow the prodn. of highly reactive radicals, such as sulfate radicals from peroxodisulfate ions, which can induce strong redox luminescence from various org. chemiluminophores including metalloporphyrins. The work presented here illustrates the generation of ECL from platinum(II) coproporphyrin (PtCP) and its bovine serum albumin (BSA) conjugate. allows the detection of these mols. below nanomolar concns. and over several orders of magnitude of concn. The relatively long. luminescence lifetime of PtCP allows discrimination from the background ECL signal using time resolved measurements, leading to higher sensitivity and the detection of PtCP-BSA indicates the potential use of metalloporphyrins as labels in ECL-based bioassays such as immunoassays and DNA-binding assays.

REFERENCE COUNT:

THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L15 ANSWER 4 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

2000:875851 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 134:187439

Dendritic Supramolecular Assembly with Multiple Ru(II) TITLE:

Tris(bipyridine) Units at the Periphery: Synthesis,

Spectroscopic, and Electrochemical Study

Zhou, Ming; Roovers, Jacques AUTHOR(S):

CORPORATE SOURCE: Institute for Chemical Process and Environmental

Technology, National Research Council of Canada,

Ottawa, ON, K1A OR6, Can.

Macromolecules (2001), 34(2), 244-252 CODEN: MAMOBX; ISSN: 0024-9297 SOURCE:

American Chemical Society

PUBLISHER: DOCUMENT TYPE: Journal

English LANGUAGE:

A supramol. assembly with eight peripheral Ru(II) tris(bipyridine), [Ru(bpy)3]2+, units covalently linked to a carbosilane dendrimer platform was synthesized. 1H NMR and MALDI-TOF mass spectrometry confirm the target structure. Spectroscopic and electrochem. studies disclose that the identical [Ru(bpy)3]2+ units in this system interact neither in the ground state nor in the excited state. In MeCN solns. of identical molar concn. of [Ru(bpy)3]2+ units, both the dendrimer and ref. monomeric [Ru(bpy)2(4-octoxy-2,2'-bipyridine)]2+ exhibit identical absorption and emission spectra. Cyclic voltammetry reveals that the dendrimer and monomer possess the same redox potentials of the metal-centered oxidn. process [Ru(bpy)3]2+/3+ and the 1st ligand-centered redn. process [Ru(bpy)3]1+/2+. The abnormal redox peaks of the [Ru(bpy)3]0/1+ and [Ru(bpy)3]1-/0 transitions of the dendrimer are attributed to the accumulation of neutral dendrimer on the electrode. A preliminary study of the electrochemiluminescence (ECL) in tripropylamine-MeCN soln. indicates that the ECL intensity of the dendrimer with eight [Ru(bpy)3]2+ units is 5 times higher than that of the ref. monomeric species. Therefore, the molar emission, generated either by photoexcitation or by electrochem. excitation, can be amplified by incorporating multiple luminophors into a multibranch platform without significantly changing the redox and photophys. properties. The possible use of supramol. assemblies as labels for biodiagnostics is discussed.

THERE ARE 77 CITED REFERENCES AVAILABLE FOR THIS 77 REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L15 ANSWER 15 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN 2000:538699 133:278254 ACCESSION NOMBER: HCAPLUS DOCUMENT NUMBER:

TITLE: Development of highly sensitive Ru-chelate based ECL

immunoassay 2: Electrochemical and immunochemical studies on homogeneous and heterogeneous ECL

excitation

Namba, Yuzaburo; Sawada, Takashi; Suzuki, Osamu AUTHOR(S): Clinical Diagnostics Department, Eisai Co., Ltd., CORPORATE SOURCE:

Tokyo, 112-8088, Japan

Analytical Sciences (2000), 16(7), 757-763 SOURCE:

CODEN: ANSCEN; ISSN: 0910-6340

Japan Society for Analytical Chemistry PUBLISHER:

DOCUMENT TYPE:

English LANGUAGE:

Studies on the electrochem. and immunochem. reaction kinetics of the heterogeneous type of ECL excitations were made comparatively with homogeneous types of ECL excitations by measuring human IgG (hIgG) using an antibody labeled with a ruthenium(II) tris(bipyridyl) (Ru-chelate) as the luminophore (Ru-Ab). Solid-phase sandwich-type immunoassays were carried out on the surface of magnetic micro-beads (MB) with a diam. of 4.5 .mu.m. In the ECL measurement, two types of ECL excitation methods were compared. One was a homogeneous ECL excitation, where the reacted MB together with non-reacted Ru-Ab were excited in a suspending phase without any bind/free (B/F) sepn. procedure. The other was a heterogeneous one where the reacted MB were excited over the electrode after being collected by a magnet following the  $\ensuremath{\text{B/F}}$  sepn. to remove the non-reacted Ru-Ab. In electrochem. studies, it was revealed that the Ru-Ab reacted with hIgG decreased the ECL emission efficiency. The decreasing ratio was inversely correlated with the cubic root of the luminophore mol. wt. In homogenous ECL excitation for the reacted matrix contg. both the reacted MB and the non-reacted Ab, however, a reverse correlating dose response curve appeared only in the area beyond the antigen-antibody equiv. point, the so-called antigen excess zone; as a result the S/N ratio of the ECL signal was as small as only 1.3. In contrast, the heterogeneous ECL excitation for the reacted MB, with the non-reacted Ru-Ab removed by B/F sepn., demonstrated 1000 or more times the S/N ratio in the area before the antigen-antibody equiv. point. Thus, this heterogeneous ECL excitation with B/F sepn. improved the detection sensitivity dramatically up to 1000 or more times higher than that of the homogeneous ECL excitation. Consequently, the sensitivity of heterogeneous ECLIA could be competitive with that of the

conventional chemiluminescence immunoassay. THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 11 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L15 ANSWER 6 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2000:382131 HCAPLUS

DOCUMENT NUMBER: 133:89174

Optical-electronic characteristics of electrolytic TITLE:

radical-recombinative luminescence

Rozhitskii, N. N. AUTHOR(S):

Khar'kovskii Gos. Tekh. Univ. Radioelektroniki, CORPORATE SOURCE:

Kharkov, 310166, Ukraine

Visnik Kharkivs'kogo Derzhavnogo Universitetu (1999), SOURCE:

440, 161-168 CODEN: VKDUF9

Kharkivs'kii Derzhavnii Universitet PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: Russian

The main optical-electronic characteristics of luminescence emitting in radical recombinative processes in electrolyzed solns. of org. compds. spectra, luminescence intensity dependencies upon electrode potential, luminophor concn., temp., effect of ultrasonic irradn. - are

discussed.

L15 ANSWER 7 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

Ceperley 09 099048 ACCESSION NUMBER: 2000:140513 HCAPLUS 132:307999 DOCUMENT NUMBER: Phenol substituent effects on electrogenerated TITLE: chemiluminescence quenching McCall, Jeff; Richter, Mark M. AUTHOR(S): Dep. Chem., Southwest Missouri State University, Springfield, MO, 65804, USA CORPORATE SOURCE: SOURCE: Analyst (Cambridge, United Kingdom) (2000), 125(3), 545-548 CODEN: ANALAO; ISSN: 0003-2654 Royal Society of Chemistry PUBLISHER: Journal DOCUMENT TYPE: LANGUAGE: English Efficient quenching of Ru(bpy)32+ (bpy = 2,2'-bipyridine) electrogenerated chemiluminescence (ECL) was obsd. in the presence of phenol and substituted phenols (e.g., 4-fluorophenol). Spectroscopic and electrochem. studies indicated that the mechanism of quenching involves energy transfer from the excited state luminophore to a benzoquinone deriv. formed at the electrode. The efficiency of ECL quenching is directly related to the position of the substituent on the arom. ring, with meta derivs. displaying the greatest magnitude of quenching. The degree of quenching does not appear to be related to the electron-donating or -withdrawing ability of the phenol substituent. THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 21 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L15 ANSWED 8 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1999:699542 HCAPLUS DOCUMENT NUMBER: 132:36150 Synthesis and characterization of partially TITLE: crosslinked poly(N-vinylcarbazole-vinyl alcohol) copolymers with polypyridyl Ru(II) luminophores. Potential materials for electroluminescence Farah, Abdiaziz A.; Pietro, William J. AUTHOR(S): Dep. Chemistry, York Univ., Toronto, ON, M3J 1P3, Can. CORPORATE SOURCE: SOURCE: Polymer Bulletin (Berlin) (1999), 43(2-3), 135-142 CODEN: POBUDR; ISSN: 0170-0839 PUBLISHER: Springer-Verlag DOCUMENT TYPE: Journal LANGUAGE: English A novel difunctionalized 5,5'-dibromomethylene-2,2'-bipyridine ligand was prepd. and covalently bound with concurrent crosslinking by a post-polymer modification method to N-vinylcarbazole-vinyl alc. copolymer. electrochem. and UV-vis spectroscopy results both confirm the covalent attachment of ruthenium transition metal complex to the polymer backbone. Differential scanning calorimetry (DSC) and thermogravimetric anal. (TGA) indicate high thermal stability of the copolymer. The copolymer is also highly phosphorescent making it a potential polymeric material for transition metal-based electroluminescent devices. THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 20 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L15 ANSWER 9 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

1998:515541 HCAPLUS ACCESSION NUMBER:

129:179853 DOCUMENT NUMBER:

TITLE: Dissolved oxygen determination by electrocatalyzed chemiluminescence with in-line solid phase media

Atwater, James E.; DeHart, Jaffrey; Wheeler, Richard AUTHOR(S):

UMPQUA Research Company, Myrtle Creek, OR, 97457, USA CORPORATE SOURCE:

Journal of Bioluminescence and Chemiluminescence SOURCE:

(1998), 13(3), 125-130

CODEN: JBCHE7; ISSN: 0884-3996

PUBLISHER: John Wiley & Sons Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Dissolved elemental O is detd. in a flowing aq. stream using glucose oxidase to catalyze the reaction between D-glucose and O to produce H2O2. The levels of the resulting H2O2 are detected and quantified by luminol chemiluminescence using in-line solid phase media for pH adjustment of the reagent stream and for controlled release of the luminophore. The reaction is initiated by electrochem. catalysis. Using excess D-glucose in the reagent flow stream, the intensity of chemiluminescence is rendered proportional only to fluctuations in the dissolved O concn. The reaction is the dissolved O concn.

rendered proportional only to fluctuations in the dissolved O conch. The detection range of aq. O is 0-10 mg/L.

REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L15 ANSWER (10) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1998:508864 HCAPLUS

DOCUMENT NUMBER: 129:146650

TITLE: Electrogenerated chemiluminescence labels for analysis

and/or referencing

INVENTOR(S): Bard, Allen J.; Richards, Thomas; Leland, Jonathan K.

PATENT ASSIGNEE(S): USA

SOURCE: U.S., 31 pp., Cont.-in-part of U.S. Ser. No. 296,630,

abandoned.
CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 3

PATENT INFORMATION:

PA'							PATENT NO. KIND DATE					APPLICATION NO. DATE										
US	US 5786141			US 5786141			Α		1998	0728		US 1995-385864 19					1995	19950209				
CA	2198																					
	2186													1996								
WO	9624																					
	W:													DK,								
		GB,	GE,	ΗU,	IS,	JP,	ΚE,	KG,	ΚP,	KR,	ΚZ,	LK,	LR,	LT,	LU,	LV,	MD,					
		MG,	MN,	MW,	MX,	NO,	NZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	ТJ,					
		TM,																				
	RW:	ΚE,	LS,	MW,	SD,	SZ,	UG,	ΑT,	ΒE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	ΙE,					
		IT,	LU,	MC,	NL,	PT,	SE,	BF,	ВJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	ML,	MR,					
				TD,																		
AU	9647	708		A.	1	1996	0827		Α	U 19	96-4	7708		1996	0209							
AU	7095	92		B	2	1999	0902											•				
EP	7554	58		A.	1	1997	0129		E	P 19	96-9	0371	6	1996	0209							
														LU,		NL,	PT,	SE				
JP	1050	2702		T	2	1998	0310		J	P 19	96-5	2429	3	1996	0209							
US	6479	233		B	1	2002	1112		Ü	S 19	98-8	2273		1998	0520							
AU	9936	767		A	1	1999	0902		A	U 19	99-3	6767		1999	0625							
PRIORIT'	Y APP	LN.	INFO	. :					US 1	994-	2966	30	В2	1994	0826							
									US 1	995-	3856	84	Al	1995	0209							
									US 1	995-	3858	64	Α	1995	0209							
									AU 1	995-	3410	3	А3	1995	0825							
									WO 1	996-	US11	13	W	1996	0209							

Biomol. anal. using anodic oxidn. of aq. sodium 9,10-diphenylanthracene-2-sulfonate (DPAS) and 1- and 2-thianthrenecarboxylic acid (1-THCOOH and 2-THCOOH) in the presence of tri-n-propylamine (TPrA) as a coreactant in aq. soln. produces electrogenerated chemiluminescence (ECL). In addn., the cathodic redn. of DPAS in the presence of peroxydisulfate (S2082-) as a coreactant also produces ECL in an acetonitrile (MeCN)-water soln. (1:1

by vol.). The oxidn. of chlorpromazine (CPZ) produces an ECL emission in the absence of an added coreactant following an unprecedented "self-annihilation" mechanism. DPAS was prepd. and the DPAS/TPrA and DPAS/S2082- ECL spectrum and emission vs. potential scans were studied. The emission spectra of DPAS at 430 nm was well sepd. from that of known luminophore tris (2,2'-bipyridine)ruthenium(II) (Ru(bpy)32+) at 620 nm.

REFERENCE COUNT: 47 THERE ARE 47 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

HCAPLUS COPYRIGHT 2003 ACS on STN L15 ANSWER 11 OF 31

ACCESSION NUMBER: 1998:455643 HCAPLUS

DOCUMENT NUMBER: 129:208495

TITLE: Hot electron injection into aqueous electrolyte

solution from thin insulating film-coated electrodes Kulmala, S.; Ala-Kleme, T.; Joela, H.; Kulmala, A. AUTHOR(S):

Department Chemistry, University Turku, Turku, CORPORATE SOURCE:

SF-20014, Finland

Journal of Radioanalytical and Nuclear Chemistry SOURCE:

(1998), 232(1-2), 91-95 CODEN: JRNCDM; ISSN: 0236-5731

Elsevier Science S.A. PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

Hot electron injection into aq. electrolyte soln. was studied with electrochemiluminescence and ESR methods. Both methods provide further indirect support for the previously proposed hot electron emission mechanisms from thin insulating film-coated electrodes to aq. electrolyte solns. The results do not rule out the possibility of hydrated electron being as a cathodic intermediate in the redn. reactions at cathodically pulse-polarized thin insulating film-coated electrodes. However, no direct evidence for **electrochem**. generation of hydrated electrons could be obtained with ESR, only spin-trapping expts. could give information about the primary cathodic steps.

L15 ANSWER 12 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1998:26670 HCAPLUS

DOCUMENT NUMBER: 128:136572

New immunoelectrochemiluminometric assay to measure TITLE:

serum thyrotropin

Sotorrio, Pilar; Quiros, Aurelia; Izquierdo, Jose AUTHOR(S):

Maria

CORPORATE SOURCE: Depto. de Bioquim. Clin., Hops. General de Asturias,

Oviedo, 33005, Spain

SOURCE: Clinical Chemistry (Washington, D. C.) (1997), 43(12),

2428-2430

CODEN: CLCHAU; ISSN: 0009-9147

PUBLISHER: American Association for Clinical Chemistry

DOCUMENT TYPE: Journal English LANGUAGE:

A new  ${\tt immunoelectrochemiluminometric}$  assay (IECMA) for the detn. of serum TSH was evaluated using the random access analyzer Elecsys 2010. The method consists of 2 immunol. "sandwich" reactions linking, at one side, sample TSH to a solid phase of paramagnetic streptavidin-coated microparticles by means of biotinylated monoclonal TSH-specific antibody and, at the other side, a Ru-labeled TSH-specific antibody. Sepn. of bound and unbound TSH is made by a magnet, and luminescence from a luminophore is triggered

electrochem. This new assay to measure TSH has an imprecision and a functional sensitivity of the third generation methods that can distinguish active hyperthyroidism from nonthyroidal diseases in accordance with recommendations of the American Thyroid Assocn. Large and small batches are conveniently analyzed and the low imprecision permits

singleton analyses.

10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L15 ANSWER (13) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

1997:459302 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 127:199423

Enzymic determination of ethanol using "reagentless" TITLE:

electrocatalyzed luminol chemiluminescence

Atwater, James E.; Akse, James R.; Dehart, Jeffrey; AUTHOR(S):

Wheeler, Richard R., Jr.

UMPQUA Research Company, Myrtle Creek, OR, 97457, USA Analytical Letters (1997), 30(8), 1445-1453 CORPORATE SOURCE:

SOURCE:

CODEN: ANALBP; ISSN: 0003-2719

PUBLISHER: Dekker DOCUMENT TYPE: Journal LANGUAGE: English

Flow anal. methodol. is presented for the detn. of aq. ethanol in concns. between 3-340 .mu.M. Alc. oxidase catalyzes the prodn. of hydrogen

peroxide which is detected by luminol chemiluminescence. Adjustment of the pH to alk. conditions and addn. of the

luminophore are implemented using in-line flow-through beds contg.

cryst. media. The requirement for a homogeneous catalyst is eliminated by

electrochem. initiation of luminescence. Traditional
lab. reagents are replaced by packed beds contg. immobilized enzyme, solid phase basification media, and cryst. luminol. The methodol. is suitable for application in both continuous flow and flow injection anal.

configurations.

L15 ANSWER (14) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

1996:742006 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 126:149655

Energetics and kinetics of homogeneous TITLE:

electrochemiluminescent electron transfer

reactions in naphthalimide solutions

Borzenko, O. V.; Bykh, A. I.; Rozhitski, N. N. AUTHOR(S):

Khar'kov State Technical Univ. Electronics, Khar'kov, CORPORATE SOURCE:

310726, Ukraine Russian Journal of Electrochemistry (Translation of

Elektrokhimiya) (1996), 32(11), 1273-1278 CODEN: RJELE3; ISSN: 1023-1935

MAIK Nauka/Interperiodica PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

Free activation energies and rate consts. for bimol. reactions

of electron transfer (RET) between electrogenerated radical ions A-. and

A+. of electrochemiluminophores A (substituted naphthalimides) are calcd. A correlation anal. is conducted for the 1st time, yielding

mechanisms and kinetics of RET that occur in an anomalous (in terms of formation of nonexcited products) region of electron transitions. The

correlation dependences of the electrochemiluminescence (ECL) intensity on the free activation energy of RET are plotted.

correlation coeffs. pertaining to each series of naphthalimide compds. under study are found for different electrolysis conditions. Comparing the above dependences and coeffs. shows adequacy of the expressions proposed for the calcn. of rate const. inherent in homogeneous RET.

calcns. are made in the framework of the known theories, describing intermol. exothermic electron transitions, which most closely coincide with the exptl. dependences of the ECL intensity on the free activation energy.

SOURCE:

L15 ANSWER (15 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

1996:741954 HCAPLUS ACCESSION NUMBER:

Ceperley 09 099048 DOCUMENT NUMBER: 126:136813 TITLE: The application of the electrochemiluminescence\*\* method for studying the mechanism of electrode \*\*\*reactions complicated by subsequent chemical stages: mechanisms of the oxidation reaction and anodic electrochemiluminescence in systems containing sodium tetraphenylborate AUTHOR(S): Belash, E. M.; Rozhitskii, N. N. Khar'kov State Technical Univ. Electronics, Khar'kov, CORPORATE SOURCE: 310726, Ukraine Russian Journal of Electrochemistry (Translation of SOURCE: Elektrokhimiya) (1996), 32(11), 1263-1272 CODEN: RJELE3; ISSN: 1023-1935 MAIK Nauka/Interperiodica PUBLISHER: DOCUMENT TYPE: Journal LANGUAGE: English The anodic reaction and homogeneous oxidn. of the Ph4B- were studied by the electrochemiluminescence (ECL) technique. Anodic ECL of a no. of electrochemiluminophores is discovered during electrolysis by a unipolar pulsed or const. voltage in a volumetric cell with ordinary electrodes. The ECL of 1,5-diphenyl-3-styrylpyrazoline in conjunction with Ph4BNa, and the calcd. parameters of the ECL diffusion kinetics demonstrate that the near-anode ECL is caused by a strongly exothermic interaction of electrogenerated radical cations with the Ph.bul. free radicals. The latter are intermediate products of homogeneous oxidn. of Ph4B- by the radical cations. L15 ANSWER (16) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1996:464529 HCAPLUS DOCUMENT NUMBER: 125:109657 Rate measurements of biomolecular reactions TITLE: using electrochemiluminescence INVENTOR(S): Nacamulli, Laurette; Leland, Jonathan K.; Hayes, Stephanie A. PATENT ASSIGNEE(S): Igen, Inc., USA SOURCE: PCT Int. Appl., 55 pp. CODEN: PIXXD2 DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND DATE	APPLICATION NO. DATE	
WO 9617248	Al 19960606	WO 1995-US15982 19951204	•
		BY, CA, CH, CN, CZ, DE, DK, EE,	
		KG, KP, KR, KZ, LK, LR, LT, LU,	
TM, TT	MW, MX, NO, NZ,	PL, PT, RO, RU, SD, SE, SG, SI,	SK, TJ,
	MW, SD, SZ, UG,	AT, BE, CH, DE, DK, ES, FR, GB,	GR, IE, O
		BF, BJ, CF, CG, CI, CM, GA, GN,	
NE SN,		110 1004 247004 10041202	
US: 5527710 CA 2206335	A 19960618 AA 19960606		
AU 9645966	A1 19960606 A1 19960619		
	B2 19991014	A0 1990 40900 19931204	
JP 10508104	T2 19980804	JP 1995-519172 19951204	
EP 871891	A1 19981021	EP 1995-944075 19951204	
R: AT, BE,	CH, DE, DK, ES,	FR, GB, GR, IT, LI, LU, NL, SE,	MC, PT, IE
PRIORITY APPLN. INFO	D.:	US 1994-347984 19941202	
		WO 1995-US15982 19951204	

AB A biomol. reaction, such as an enzymic or affinity binding reaction, is effected in an electrochem

4

. cell with a mixt. of reagents including a luminophor which will relate the concn. of a reactant, a reaction partner, or the reaction product of a reaction partner to the electrochemiluminescent intensity to det. the rate of the biomol. reaction. The reaction partner is a reagent which reacts with the reactant and which itself or its reaction product participates with the luminophor to cause ECL emissions. The ECL intensity is modulated with a series of elec. pulses which are applied to the mixt. of reagents at a preselected potential for preselected intervals of time and duration. The intensity is measured during those intervals to provide a timed series of values (P). The same expt. is repeated twice more, once wherein the modulation is conducted after the reaction has gone to completion to obtain the timed series of values (C), and the last time the reaction is performed in the absence of the reaction partner to obtain the timed series of values (B). The results are normalized (N) using the formula N = (P - B)/(C - B) to obtain a series of values (N) which are plotted to obtain the time course of reaction.

L15 ANSWER 17 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

Journal

ACCESSION NUMBER: 1994:425392 HCAPLUS

DOCUMENT NUMBER:

121:25392 Synthesis and intramolecular energy transfer of the TITLE:

AUTHOR(S): CORPORATE SOURCE:

ruthenium-ferrocene polynuclear complexes

Duan, Chunying; You, Xiaozeng; Tao, Hui; Zhu, Longgen Coord. Chem. Inst., Nanjing Univ., Nanjing, 210008,

Peop. Rep. China

Gaodeng Xuexiao Huaxue Xuebao (1993), 14(11), 1492-7 SOURCE:

CODEN: KTHPDM; ISSN: 0251-0790

DOCUMENT TYPE:

Chinese LANGUAGE:

GT

$$N = N = 0$$

$$N = 0$$

[L2Ru(bpy)](PF6)2 (L = I; R = H, Me, n = 0-2) and [FeL3](PF6)2 were prepd. AB and characterized. The luminescence of the chromophore Ru(II) is quenched by the luminophor Fe(II), and efficient intramol. energy transfer from the chromophore Ru(II) to the luminophor Fe(II)occurs in these bimetallic complexes. The efficiency of the intramol. energy transfer is related to the no. of the luminophor Fe(II) around the chromophore Ru(II) and the distance between the Ru(II) and Fe(II) in the bimetallic complexes.

L15 ANSWER (18 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

1992:50213 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 116:50213

Intramolecular energy transfer in ruthenium(II)-TITLE:

chromium(III) chromophore-luminophore

complexes. Ru(bpy)2[Cr(cyclam)(CN)2]24+ Bignozzi, Carlo Alberto; Bortolini, Olga; Chiorboli, AUTHOR(S): Claudio; Indelli, Maria Teresa; Rampi, Maria Anita; Scandola, Franco CORPORATE SOURCE: Dip. Chim., Univ. Ferrara, Ferrara, 44100, Italy Inorganic Chemistry (1992), 31(2), 172-7 SOURCE: CODEN: INOCAJ; ISSN: 0020-1669 DOCUMENT TYPE: LANGUAGE: English New chromophore-lumophor RuII(bpy)2[CrIII(cyclam)(CN)2]24+ (bpy = 2,2'-bipyridine) was synthesized and characterized. Visible light absorption by the Ru(bpy)22+ chromophore leads to emission from the Cr(cyclam) (CN) 2+ luminophor, as a consequence of efficient (.qtoreq.99%) and fast (sub-ns time scale) chromophore-luminophor exchange energy-transfer process. The emission is intense (.PHI. = 5.3 .times. 10-3 in H2O) and long-lived (.tau. = 260 .mu.s in H2O). photophys. properties of the **luminophor** are slightly perturbed by interaction with the chromophore, resulting in a sharper emission band shape and shorter radiative and radiationless lifetimes. The presence of a  ${\tt Ru}({\tt II})$  .fwdarw.  ${\tt Cr}({\tt III})$  intervalence transfer state, hardly detectable in the ground-state spectrum, is revealed by the excited-state absorption spectrum of the chromophore-luminophor complex. L15 ANSWER (19) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1990:640046 HCAPLUS DOCUMENT NUMBER: 113:240046 TITLE: Electrogenerated chemiluminescence: An oxidative-reduction type ECL reaction sequence using tripropyl amine AUTHOR(S): Leland, Jonathan K.; Powell, Michael J. CORPORATE SOURCE: IGEN Inc., Rockville, MD, 20852, USA Journal of the Electrochemical Society (1990), SOURCE: 137(10), 3127-31 CODEN: JESOAN; ISSN: 0013-4651 DOCUMENT TYPE: Journal LANGUAGE: English A new electrogenerated chemiluminescence (ECL) reaction with utilizes tri-Pr amine and Ru(bpy)32+ is presented. The mechanism of light generation appears general enough to include a range of amines and luminophors. An oxidative-redn. mechanism is proposed. Upon electrochem. oxidn. of both the luminophor and amine, a strong emission is obsd. Voltammetric anal. reveals the potential for greatest light emission at the tri-Pr amine oxidn. The emission is from the excited state of Ru(bpy)32+. An electron transfer reaction from the deprotonated tri-Pr amine radical and Ru(bpy) 33+ is the central reaction for excited state prodn. An est. of the ECL efficiency cannot be made owing to the complex nature of the reaction. L15 ANSWER 20 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1988:642953 HCAPLUS DOCUMENT NUMBER: 109:242953 Synthetic routes to new polypyridyl complexes of TITLE: osmium(II) Kober, Edward M.; Caspar, Jonathan V.; Sullivan, B. AUTHOR(S): Patrick; Meyer, Thomas J. CORPORATE SOURCE: Dep. Chem., Univ. North Carolina, Chapel Hill, NC, 27514, USA SOURCE: Inorganic Chemistry (1988), 27(25), 4587-98 CODEN: INOCAJ; ISSN: 0020-1669 DOCUMENT TYPE: Journal LANGUAGE: English New luminescent complexes of Os(II) that contain either 2,2'-bipyridine (bpy) or 1,10-phenanthroline (phen) as the chromophoric acceptor ligand

were prepd. by a combination of established and new synthetic methods. Extenstive use of Os(IV) and Os(III) precursors , e.g., OsIV(bpy)Cl4 and mer-OsIII(PMe2Ph)Cl3, led to the prepn. of materials with ancillary ligands such as tertiary phosphines as preparative intermediates, including OsIII (bpy) (PMe2PH) Cl3 and cis-OsII (phen) LCl2 (L = phosphines). Further substitution of Cl- into complexes such as these gave emissive Os(II) complexes. Another new synthetic route utilizes the versatile Os(II) percursor Os(bpy)2CO3, which allows the facile prepn. of dicationic, disubstituted species such as [Os(bpy)2(nor)]2+ (nor = norbornadiene). Another general procedure, based on the control of solvent and temp. in the substitution chem. of cis-Os(bpy)2Cl2, was further developed to produce a variety of cis- [Os(bpy)2L1Cl]+ and cis-[Os(bpy)2L12]2+ (L1 = phosphine, arsine, N-donor ligand or olefin). In a few cases, phosphine entering groups cause the cis geometry to be unfavorable and new trans-[Os(bpy)2L2]2+ were isolated. Because a major emphasis of this work was to develop a wide variety of luminophors bases on OsII-bpy polypyridine type chromophores, the resultant complexes comprise the largest family of transition-metal-bases excited-state reagents with tunable photophys. and redox properties available. When possible, the new complexes were characterized by UV-visible and emission spectroscopy, cyclic voltammetry and 31P and/or 1H NMR spectroscopy.

L15 ANSWER (21) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN 1988:578804 HCAPLUS

ACCESSION NUMBER: DOCUMENT NUMBER: 109:178804

TITLE: Evidence for a radical intermediate in the anodic

oxidation of reduced nicotinamide adenine dinucleotides obtained by electrogenerated

chemiluminescence Ludvik, J.; Volke, J.

AUTHOR(S): J. Heyrovsky Inst. Phys. Chem. Electrochem., Czech. CORPORATE SOURCE:

Acad. Sci., Prague, 182 23, Czech.

SOURCE: Analytica Chimica Acta (1988), 209(1-2), 69-78

CODEN: ACACAM; ISSN: 0003-2670

DOCUMENT TYPE: Journal LANGUAGE: English

Electrogenerated chemiluminescence is used to show that the radicals NAD.bul. and NADP.bul. are intermediates in the electrooxidn. of NADH and NADPH at a Pt anode in anhyd. or partly aq. (up to 15 vol.%) DMSO. An ECE mechanism seems to predominate. The use of DMSO proved to be very convenient, with the advantage of enabling electrogenerated chemiluminescence to be obtained in partly aq. media even with ionic substances as substrates. The method is useful in proving the existence of unstable radical intermediates in redox processes, even for relatively large mols. such as NADH and NADPH.

L15 ANSWER 22 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

1987:563894 HCAPLUS ACCESSION NOMBER:

DOCUMENT NUMBER: 107:163894

Mechanism of electrochemiluminescence of TITLE:

chloride ion-containing compositions Bykh, A. I.; Kukoba, A. V.; Rozhitskii, N. N.

AUTHOR(S): Khar'k. Inst. Radioelektron., Kharkov, USSR CORPORATE SOURCE:

Elektrokhimiya (1987), 23(7), 928-35 SOURCE:

CODEN: ELKKAX; ISSN: 0424-8570 Journal

DOCUMENT TYPE: LANGUAGE: Russian

The redn. and oxidn. of some organoluminophors were investigated by using voltammetry with simultaneous recording of electrochemiluminescence. In aprotic solvents contg. Cl-, the increase of the intensity of the chemiluminescence is explained by the partitioning of Cl- in the homogeneous reaction with cation radicals (A+.bul.) and formation of new radicals ACl.bul..

reaction of electron transfer between AlCl.bul. and electrogenerated anion radicals is responsible for the chemiluminescence signal due to the deactivation of the 1A.beta.ull mols.

L15 ANSWER (23) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1987:127768 HCAPLUS

DOCUMENT NUMBER: 106:127768

TITLE: investigation of two radical intermediates in the

anodic oxidation of 1,4-dihydropyridines by

electrochemiluminescence

AUTHOR(S): Ludvik, Jiri; Volke, Jiri; Pragst, Fritz

CORPORATE SOURCE: J. Heyrovsky Inst. Phys. Chem. Electrochem., Czechoslovak Acad. Sci., Prague, 118 40, Czech.

SOURCE: Journal of Electroanalytical Chemistry and Interfacial

Electrochemistry (1986), 215(1-2), 179-90

CODEN: JEIEBC; ISSN: 0022-0728

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The chemiluminescence generated in the anodic oxidn. of

1,4-dihydropyridines (PyRH) in the presence of 9,10-diphenylanthracene
(DPA) [1499-10-1] or of bis-[1,2,3-trimethyl-1,2-dihydrobenzimidazolyl(2)] [94887-83-9] luminophor systems was investigated at a
rotating Pt disk electrode and by cyclic voltammetry connected with
luminescence intensity measurements in a 1:1 MeCN + toluene mixt.

New findings on the electroluminescence of PyRH are reported;
the cation radicals PyRH.bul.+ are the primary intermediates. The snape
and the intensity of the luminescence-potential curves are
discussed in terms of structure. If El/2ox(PyRH) < El/2ox(DPA), the
system PyRH/DPA is a practical example of the homogeneous redox catalysis

L15 ANSWER (24) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1987:77689 HCAPLUS

DOCUMENT NUMBER: 106:77689

of electrochem. reactions.

TITLE: Structure and redox and photophysical properties of a

series of ruthenium heterocycles based on the ligand

2,3-bis(2-pyridyl)quinoxaline

AUTHOR(S): Rillema, D. Paul; Taghdiri, Donna G.; Jones, Daniel

S.; Worl, Laura A.; Meyer, Thomas J.; Levy, Henri A.;

Keller, Charles D.

CORPORATE SOURCE: Dep. Chem., Univ. North Carolina, Charlotte, NC,

28223, USA

SOURCE: Inorganic Chemistry (1987), 26(4), 578-85

CODEN: INOCAJ; ISSN: 0020-1669

DOCUMENT TYPE: Journal LANGUAGE: English

The prepn., structure, and redox and photophys. properties of [Ru(bpy)n(BL)3-n]2+ (n = 0-2; bpy = 2,2'-bipyridine; BL = 2,3-bis(2-pyridyl)quinoxaline) are described. [Ru(bpy)2(BL)](PF6)2 crystd. in the monoclinic space group P21/a with a 14.664(4), b 16.345(5), c 18.978(5) .ANG., .beta. 100.19(2).degree., Z = 4, R = 0.0672 for 5766 reflections with I > .sigma.(I). The N-Ru bond distances were Ru-N(pyridine) = 2.06 .ANG. (av.) and Ru-N(pyrazine) = 2.096(4) .ANG.. Absorption spectra contained bands (250-300 nm) in the UV region that were assignable to ligand .pi. .fwdarw. .pi.\* transitions and visible bands (517-300 nm) that were assignable to d.pi. .fwdarw. .pi.\* MLCT transitions. Redn. potentials for the Ru3+/2+ couples were 1.40-1.70 V vs. SSCE. Three redns. were obsd. and assigned to the 1-electron redn. of each bidentate ligand, commencing with BL and then followed by bpy. Weak luminescence (.PHI.r = 0.012-0.002 in 4:1 EtOH-MeOH) was obsd., and cor. emission energy max. were located at 766 .+-. 4 nm. Excited-state redn. potentials were estd. from the difference between emission energy max. and





ground-state redn. potentials. Potentials for excited-state Ru2+\*/+ couples were detd. Excited-state lifetimes at room temp. in 4:1 EtOH-MeOH were 167 to <70 ns. Temp.-dependent-lifetime (90-298 K) data gave evidence for .DELTA.E' values of 1100 .+-. 300 cm-1. Emission spectral fitting suggested contributions to the spectral profile from medium-energy ring stretching modes and low-energy metal-ligand stretching vibrations. A plot of E00 vs. .DELTA.E1/2 for a series of Ru luminophores, where .DELTA.E1/2 = E1/23+/2+ - E1/22+/+, was linear with a slope of 0.87 and a correlation coeff. of 0.97.

L15 ANSWER (25)OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1986:236607 HCAPLUS

DOCUMENT NUMBER: 104:236607

TITLE: Electrogenerated chemiluminescence in mechanistic

investigations of electroorganic reactions.

Part VI. Sensitive detection of cation radicals by bis[1,2,3-trimethyl-2,3-dihydrobenzimidazolyl-(2)]/

luminophor systems

AUTHOR(S): Pragst, F.; Niazymbetov, M.

CORPORATE SOURCE: Sekt. Chem., Humboldt-Univ., Berlin, DDR-1040, Ger.

Dem. Rep.

SOURCE: Journal of Electroanalytical Chemistry and Interfacial

Electrochemistry (1986), 197(1-2), 245-64

CODEN: JEIEBC; ISSN: 0022-0728

DOCUMENT TYPE: Journal LANGUAGE: English

Cation radicals D.+, formed as the primary products in an anodic process, produce a luminescence signal in the potential region of the voltammetric wave of D, if the oxidn. is carried out in the presence of bis[1,2,3-trimethyl-2,3-dihydrobenzimidazolyl-(2)], B2, and a suitable luminophor, A, such as 9,10-diphenylanthracene (DPA), 2-phenyl-4-p-biphenylyl-1,3,4-oxadiazole (PBD) or other highly fluorescent arom. hydrocarbons, oxazoles or oxadiazoles. The use of this luminescence signal for a sensitive detection of D.+ was tested at a rotating disk electrode in a 1:1 MeCN + PhMe mixt. as the solvent for a series of arom. and aliph. amines, .DELTA.2-pyrazolines, carbazole, indole, 1,4-dihydropyridines, phenols, methoxybenzenes and tri-arylphosphines. For stable cation radicals the luminescence-potential curve shows a prewave at the foot of the voltammetric wave of D and a luminescence plateau in the limiting current region. In the case of a chem. reaction of D.+ the plateau is diminished or vanishes completely and the prewave changes to a prepeak between 20 and 100 mV below  $\mathrm{E1/2ox}(\mathrm{D})$  . The prepeak also decreases with increasing rate of the subsequent reaction, but is still seen for very short lived D.+, e.g. for hydroquinone, carbazole or triphenylphosphine. The advantages and the restrictions of the method are discussed in the context of the luminescence mechanism.

L15 ANSWER (26) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1977:497777 HCAPLUS

DOCUMENT NUMBER: 87:97777

TITLE: Modeling of neural mechanisms in the

electrochemiluminescent cell. Communication

2. Adaptation

AUTHOR(S): Leonov, V. P...

CORPORATE SOURCE: USSR

SOURCE: Problemy Bioniki (1976), 16, 136-40

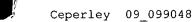
CODEN: PBNKAV; ISSN: 0555-2656

DOCUMENT TYPE: Journal LANGUAGE: Russian

AB The neuronal adaptation mechanism was mimicked in

electrochemiluminescent cells (electrochem. cells contg.

luminophores in their electrolyte solns.) in which the working





electrode was connected to 1 terminal of a square-wave impulse generator, a 2nd electrode was connected through a condenser to the other terminal of the generator, and 2 other electrodes were connected to a source of const. polarizing potential. The nerve impulse transmission, represented by electrochemiluminescence, was obsd. under appropriate conditions in solns. contg. 1,5-diphenyl-3-styrylpyrazoline (I) plus LiCl (as electrolyte) or Na tetraphenylborate plus I. The mechanisms for these electrochemiluminescent reactions are discussed. With increasing potential of the impulse on the working electrode above a threshold value, the electrochemiluminescent intensity (synaptic transmission) increased to a satn. value. With increasing polarizing potential, the luminescence intensity at a given impulse potential increased and the satn. value for luminescence was also increased. After the polarizing potential was turned off, luminescence initially increased and then slowly decreased. Thus, these systems mimic the general phenomena obsd. in neuronal adaptation.

L15 ANSWER 27-OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1977:497776 HCAPLUS

DOCUMENT NUMBER: 87:97776

TITLE: Modeling of neural mechanisms in the

electrochemiluminescent cells. Communication

I

AUTHOR(S): Leonov, V. P.

CORPORATE SOURCE: USSR

SOURCE: Problemy Bioniki (1976), 16, 99-104

CODEN: PBNKAV; ISSN: 0555-2656

DOCUMENT TYPE: Journal LANGUAGE: Russian

Electrochemiluminescent cells (electrochem. cells contq. luminophores in their electrolyte solns.) were analyzed as possible models for synaptic excitation, inhibition, adaptation, and summation. In electrochem. cells contg. 1,5-diphenyl-3styrylpyrazoline (I), luminescence (synaptic excitation) was obsd. when a I-cation radical reacted with a I-anion radical to form a luminescent excited mol. or when a 2nd electron was transferred to the I-cation radical. The anodic reaction representing the presynaptic input can be used to model summation (facilitation) with the rate of the reaction modeling spatial summation and the life-time of the I-cation radical modeling temporal summation. Control of the rate of the cathodic reaction mimics adaptation. electrochem. cells contg. Na tetraphenylborate plus triphenylimidazole, 1,5-diphenyl-3-(p-chlorophenyl)pyrazoline, or dibenzyl, electrochemiluminescence was obsd. as the result of the formation of exciplexes between the tetraphenylborate anion and the cation radicals of the other compds. This system can mimic the same synaptic mechanism as the system contg. I. In addn., if a neg. impulse is applied to the synaptic electrode of this 2nd system before the pos.

L15 ANSWER (28) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1975:599692 HCAPLUS

can also be modeled in this system.

DOCUMENT NUMBER: 83:199692

TITLE: Kinetics of aging of electroluminophors
AUTHOR(S): Matisen, L.; Talviste, E.; Tammik, A.

impulse, luminescence is inhibited. Thus, synaptic inhibition

CORPORATE SOURCE: USSR

SOURCE: Tartu Riikliku Ulikooli Toimetised (1975), 346, 132-41

CODEN: UTGUA3; ISSN: 0494-7304

DOCUMENT TYPE: Journal LANGUAGE: Russian

AB The aging of an EL-516 electroluminophor (also a surface-treated sample was examd.) in different relative humidity (over satd. solns. of



P205, LiCl, CaCl2, or NaBr) of the working medium for electroluminescence capacitors (prepd. by a dry dusting of the electroluminophor powder), was studied. The supposition about a 2-stage aging was confirmed; in the initial stage, the aging kinetics were mainly bimol., changing later (mostly after the brightness drop to <50%) into a distinct monomol. kinetics with a const. aging rate, independent of humidity. The aging kinetics can be used for predicting the electroluminophor service lifetime; the brightness level at which the change of the kinetics occurs can serve an indicator of the luminophor stability perfection. The possible aging reactions are discussed. The surface coating (K. V. Realo, et el., 1972) resulted in a shortening of initial aging stages (braked the electrochem. decompn. of luminophors) and a decrease of aging rate in the stage of monomol. kinetics (a decrease of impurity and vacancy diffusion).

L15 ANSWER 29 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1974:75550 HCAPLUS

DOCUMENT NUMBER: 80:75550

TITLE: Stability of powdered electroluminophors

with protective electrochemical coatings AUTHOR(S): Soshchin, N. P.; Talviste, E.; Tammik, A.

CORPORATE SOURCE: USSR

SOURCE: Tartu Riikliku Ulikooli Toimetised (1972), No. 292,

13-25

CODEN: UTGUA3; ISSN: 0494-7304

DOCUMENT TYPE: Journal LANGUAGE: Russian

AB The formation of protective layers on the surface of powd. ZnS
electroluminophors led to some modifications of the parameters of
the centers responsible for fast and slow decay of brightness. Coatings
from compds. more electropos. than ZnS had the greatest effect on the
decay processes. This agrees with the theory of electrochem.
treatment of electroluminescent layer aging processes. This
study showed the possibility of application of consecutive
electrochem. reactions in electroluminescent
capacitors.

L15 ANSWER 30 OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1972:133008 HCAPLUS

DOCUMENT NUMBER: 76:133008

TITLE: Electrochemical nature of electroluminophor

aging

AUTHOR(S): Soshchin, N. P.; Orlov, I. N.

CORPORATE SOURCE: USSR

SOURCE: Elektrolyumin. Tverd. Tel, Tr. Soveshch.

Elektrolyumin., 3rd (1971), Meeting Date 1969, 279-83. Editor(s): Gol'dman, A. G. "Naukova Dumka": Kiev,

Ukr. SSR. CODEN: 24PJAQ

DOCUMENT TYPE: Conference LANGUAGE: Russian

AB The life of luminophors based on ZnS.Cu2S depends on the amt. of Cu, which decreases due to the electrochem. oxidn. of Cu2S to CuS. This diminishes the luminance of the luminophor. Also, the presence of CuS decreases the cond. of the electroluminescent layer. Some decrease of the emf. of an electrolytic cell formed in the

layer. Some decrease of the emf. of an electrolytic cell formed in this layer can be attained by an introduction of more pos. compds., such as CdS or ZnSe, into the base of the luminophor.

L15 ANSWER (3) OF 31 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NOWHER: 1971:545801 HCAPLUS

DOCUMENT NUMBER: 75:145801

Generation of luminophors by the action of

TITLE:



```
an electrical current on a sodium chloride melt and
                         their optical properties
AUTHOR(S):
                         Zemlicka, J.; Dolejsi, J.
CORPORATE SOURCE:
                         Inst. Solid State Phys., Czech. Acad. Sci., Prague,
                         Czech.
SOURCE:
                         Kristall und Technik (1971), 6(2), K7-K11
                         CODEN: KRTEAW; ISSN: 0023-4753
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         German
     By using chem. reactions that take place in a NaCl melt during
     electrochem. processes, optical, and particularly
     luminescent properties of NaCl crystals, are influenced. NaCl + M
     (M = Cu, Au, Fe, and Ni) luminophors were obtained by the action
     of d.c. on a NaCl melt. The O2- ion is generated in the melt. This was
     identified from the spectral compn. of the fluorescence. This ion is
     responsible for the yellow fluorescence. The green and orange-yellow
     fluorescence is ascribed to exochloro complexes. The O-Cl ligands that
     surround the central metal ions are assigned greater importance than the
     metal ion.
=>
=>
=> d stat que 116
              1 SEA FILE=REGISTRY ABB=ON PLU=ON ELECTROCHEM/BI
             13 SEA FILE=REGISTRY ABB=ON
                                          PLU=ON
                                                  LUMINOPHOR/BI
L_2
L3
              1 SEA FILE=REGISTRY ABB=ON
                                          PLU=ON
                                                  LUMINOPHORE/BI
                                         PLU=ON LUMINESC?
          13142 SEA FILE=REGISTRY ABB=ON
L7
         275185 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L1 OR ?ELECTROCHEM?
1.8
         230077 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L7 OR ?LUMINESC?
           4251 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L8(L)L9
L10
L11
                SEL PLU=ON L3 1- CHEM:
                                                2 TERMS
L12
              1 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L11
                                         PLU=ON L12 OR L2 OR ?LUMINOPHOR?
1.13
           3990 SEA FILE=HCAPLUS ABB=ON
             48 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L10 AND L13
T.14
L15
             31 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L14 AND (REACTION OR COURSE
                OR BINDING)
             17 SEA FILE=HCAPLUS ABB=ON PLU=ON L14 NOT L15
L16
=>
=>
=> d ibib abs hitrn 116 1-17
L16 ANSWER 1 OF 17
                     HCAPLUS COPYRIGHT 2003 ACS on STN
ACCESSION NUMBER:
                         2002:535693 HCAPLUS
DOCUMENT NUMBER:
                         137:301224
TITLE:
                         Electrochemiluminoimmunoassay of hTSH at disposable
                         oxide-coated n-silicon electrodes
                         Helin, M.; Vare, L.; Hakansson, M.; Canty, P.; Hedman,
AUTHOR(S):
                         H.-P.; Heikkila, L.; Ala-Kleme, T.; Kankare, J.;
                         Kulmala, S.
                         Laboratory of Inorganic and Analytical Chemistry,
CORPORATE SOURCE:
                         Helsinki University of Technology, Helsinki, Hut,
                         02015, Finland
SOURCE:
                         Journal of Electroanalytical Chemistry (2002),
                         524-525, 176-183
                         CODEN: JECHES
                         Elsevier Science B.V.
PUBLISHER:
DOCUMENT TYPE:
                         Journal
                         English
LANGUAGE:
```



AΒ Cathodic pulse polarization of thin insulating film-coated electrodes enables the generation of electrochemiluminescence (ECL) by tunnel emission of hot electrons from the Fermi level of the conductor material of the conductor-insulator-aq. electrolyte soln. junction to the solutes at the vicinity of the electrode surface and probably also to the conduction band of water. The latter process can generate hydrated electrons as strongly reducing slightly longer-lived cathodic intermediates, which are known to be able to induce chemiluminescence (CL) of various types of luminophors having very different photophys. and chem. properties. The generation of the above-mentioned cathodic primary species provides good possibilities to use many types of luminophors as label mols. in sensitive immuno and DNA-probing assay. This paper introduces an electrochemiluminoimmunoassay (ECLIA) for human TSH (hTSH) at oxide-coated n-silicon electrodes and demonstrates the suitability of silicon electrodes covered with thermally grown silicon dioxide film as disposable working electrodes (WEs) in sensitive time-resolved ECL (tr-ECL) measurements in aq. soln. The label chelate can be detected almost down to picomolar level and the calibration curve of the chelate covers more than five orders of magnitude of chelate concn. Also the calibration curve of the immunometric hTSH assay was found to be linear over a wide range of hTSH concn., the detection limit of the hormone being below 1 mU 1-1 (4 pmol 1-1).

REFERENCE COUNT: 34 THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 2 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2002:149842 HCAPLUS

DOCUMENT NUMBER: 136:352197

TITLE: Hot electron-induced electrogenerated

chemiluminescence of 1-aminonaphthalene-4-sulphonate

at oxide-covered aluminium electrodes in aqueous

solution

AUTHOR(S): Helin, M.; Hakansson, M.; Canty, P.; Spehar, A.-M.;

Kulmala, S.

CORPORATE SOURCE: Department of Inorganic and Analytical Chemistry,

Helsinki University of Technology, HUT, Helsinki,

FIN-00215, Finland

SOURCE: Analytica Chimica Acta (2002), 454(2), 193-201

CODEN: ACACAM; ISSN: 0003-2670

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

Electrogenerated chemiluminescence of 1-aminonaphthalene-4-sulfonate (ANS) provides a sensitive means for the detection of the ANS in aq. soln. when oxide-covered aluminum electrodes are used as tunnel emission electrodes and cathodic pulse polarization is used for the excitation of the luminophore. During the pulse polarization of insulating oxide film-coated aluminum cathodes, hot electrons are tunnel emitted from the aluminum to the aq. electrolyte soln. by direct field-assisted tunneling (in the case of oxide films of thickness 2-6 nm) or by Fowler-Nordheim (FN) tunneling in the cases of thicker oxide films. As a result of direct tunnel emission of these energetic electrons, the generation of hydrated electrons (eaq-) becomes possible. These electrochem. generated, extremely strong reductants (dry hot or hydrated electrons) make the efficient excitation of various types of luminophores at thin insulating film-covered electrodes possible and provide a means for sensitive immunoassays and DNA-probing assays when these luminophores are used as label mols.

REFERENCE COUNT: 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER (3) OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

2001:397155 HCAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: 135:13603 Thin film fiber optic electrode sensor array and apparatus Pantano, Paul; Jin, Eunsook S.; Khan, Samina S.; INVENTOR(S): Stokes, Harold W., Jr. Board of Regents the University of Texas System, USA PATENT ASSIGNEE(S): PCT Int. Appl., 69 pp. SOURCE: CODEN: PIXXD2 DOCUMENT TYPE: Patent English LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: APPLICATION NO. DATE PATENT NO. KIND DATE WO 2001038853 WO 2000-US32478 20001129 A2 20010531 A3 20020103 WO 2001038853 AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG US 6487326 В1 20021126 US 1999-450156 19991129 A 19991129 US 1999-450156 PRIORITY APPLN. INFO.: An app. and method is disclosed for detecting an analyte using a fiber optic electrochem. sensor that includes a fiber optic layer, an elec. conductive translucent metallic layer and a light energy absorbing dye layer. L16 ANSWER/4 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 2000:795963 HCAPLUS TITLE: Synthesis and characterization of d5 Re(II) diphosphine luminophores. Enyeart, Stephania J.; Kirchhoff, Jon R. AUTHOR(S): Department of Chemistry, University of Toledo, Toledo, CORPORATE SOURCE: OH, 43606, USA Abstracts of Papers - American Chemical Society SOURCE: (2000), 220th, INOR-515 CODEN: ACSRAL; ISSN: 0065-7727 American Chemical Society PUBLISHER: Journal; Meeting Abstract DOCUMENT TYPE: LANGUAGE: English A series of rhenium complexes with the general formulas [Re(PP)3]n+ and [Re(PP)2(PP')]n+, where PP and PP' are either 1,2bis(diethylphosphino)ethane (depe) or 1,2-bis(dimethylphosphino)ethane (dmpe) and n=1 or 2, were synthesized and characterized. A new synthetic method to low valent Re(I) and Re(II) complexes was developed involving a redn./substitution approach starting with a Re(III) precursor in 1,2-dimethoxyethane. Depending on the combination of the ligand donating properties, either the Re(I) or Re(II) complex was prepd. While the Re(I)complexes do not luminesce in room temp. soln., the Re(II) complexes are rare examples of highly luminescent d5 transition metal complexes. The series was characterized by absorption and emission spectroscopy, electrochem., and single crystal X-ray anal. In

situ conversion of the Re(I) complexes to the Re(II) analogs by

characterization of the Re(II) excited state for all complexes. Correlations between the structural properties and the **electrochem** 

luminescence spectroelectrochem. permitted

. and spectroscopic properties will be discussed.

L16 ANSWER 5 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1999:300532 HCAPLUS

DOCUMENT NUMBER: 131:65187

Quenching of Electrogenerated Chemiluminescence by TITLE:

Phenols, Hydroquinones, Catechols, and Benzoquinones

McCall, J.; Alexander, C.; Richter, M. M. AUTHOR(S):

Department of Chemistry, Southwest Missouri State University, Springfield, MO, 65804, USA Analytical Chemistry (1999), 71(13), 2523-2527 CODEN: ANCHAM; ISSN: 0003-2700 CORPORATE SOURCE:

SOURCE:

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

Efficient quenching of Ru(bpy)32+ (bpy = 2,2'-bipyridine) electrogenerated chemiluminescence was obsd. in the presence of phenols, catechols, hydroquinones, and benzoquinones. In most instances, quenching is obsd.

with 100-fold excess of quencher over Ru(bpy)32+, with complete quenching obsd. at 1000-2000-fold excess. The mechanism of quenching is believed to involve energy transfer from the excited-state luminophor to

benzoquinone. In the case of phenols, catechols, and hydroquinones, quenching is believed to occur via a benzoquinone deriv. formed at the

electrode surface. Luminescence and UV-visible expts. coupled with bulk electrolysis support the formation of benzoquinone products upon electrochem. oxidn.

REFERENCE COUNT: 43 THERE ARE 43 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 6 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1999:196541 HCAPLUS

DOCUMENT NUMBER: 130:290707

TITLE: Electrochemiluminescent labels for

applications in fully aqueous solutions at

oxide-covered aluminum electrodes

Kulmala, S.; Helin, M.; Ala-Kleme, T.; Vare, L.; AUTHOR(S):

Papkovsky, D.; Korpela, T.; Kulmala, A.

CORPORATE SOURCE: PO Box 6100, Department of Chemical Technology,

Laboratory of Inorganic and Analytical Chemistry, Helsinki University of Technology, FIN-02015 HUT,

Helsinki, Finland

Analytica Chimica Acta (1999), 386(1-2), 1-6 SOURCE:

CODEN: ACACAM; ISSN: 0003-2670

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

Oxide-covered aluminum electrodes as well as other tunnel emission electrodes allow various label mols. having very different redox and optical properties to be excited cathodically. Low detection limits are obtained and the linear calibration concn. range of the labels spans 5 or 6 orders of magnitude. The lowest detection limits are obtained with Tb(III) chelates which can be detected down to pico-molar levels in aq. soln. using time-resolved measurement techniques. Luminophores, such as, 9-fluorenylmethyl chloroformate, derivs. of fluorescein and its analogs, arom. lanthanide(III) chelates, various coumarins and porphyrins can be used as labels emitting in different spectral regions. extraordinary anal. power of the tunnel emission electrodes lies in the possibility of simultaneously exciting several different labels emitting either in the UV, visible or NIR range and luminescence lifetimes varying from the ns to the ms range. Therefore, wavelength or

time discrimination or their combination can be exploited in sepn. of the electrochemiluminescence signals from different labels.

THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 19



RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

bF 17 HCAPLUS COPYRIGHT 2003 ACS on STN L16 ANSWER

1999:143857 HCAPLUS ACCESSION NUMBER:

Quenching of electrochemiluminescent excited TITLE:

states by phenols, catechols, benzoquinones and

Richter, M. M.; McCall, J.; Alexander, C. AUTHOR(S): CORPORATE SOURCE: Dept. of Chemistry, Southwest Missouri State

University, Springfield, MO, 65804, USA

Book of Abstracts, 217th ACS National Meeting, SOURCE: Anaheim, Calif., March 21-25 (1999), ANYL-021.

American Chemical Society: Washington, D. C.

CODEN: 67GHA6

DOCUMENT TYPE: Conference; Meeting Abstract

LANGUAGE: English

Efficient quenching of Ru(bpy) 32+ (bpy = 2,2'-bipyridine) electrogenerated chemiluminescence (ECL) was obsd. in the presence of phenols, catechols, hydroquinones and benzoquinones. Spectroscopic and electrochem. studies indicate that the mechanism of quenching involves energy transfer from the excited state luminophore to benzoquinone; or in the case of phenols, hydroquinones and catechols, a benzoquinone deriv. formed directly at the electrode. Results to date as

well as potential applications will be discussed. L16 ANSWER (8) DF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1999:143856 HCAPLUS TITLE:

Phenol substituent effects on the quenching of

electrogenerated chemiluminescent excited states

AUTHOR(S):

McCall, J.; Richter, M. M. Dept. of Chemistry, Southwest Missouri State CORPORATE SOURCE:

University, Springfield, MO, 65804, USA Book of Abstracts, 217th ACS National Meeting,

Anaheim, Calif., March 21-25 (1999), ANYL-020. American Chemical Society: Washington, D. C.

CODEN: 67GHA6

Conference; Meeting Abstract DOCUMENT TYPE:

LANGUAGE: English

SOURCE:

AUTHOR(S):

Efficient quenching of Ru(bpy)32+(bpy = 2,2'-bipyridine) electrogenerated chemiluminescence (ECL) has been obsd. in the presence of phenol and substituted phenols (e.g., 4-fluorophenol). Spectroscopic and electrochem. studies indicate that the mechanism of quenching involves energy transfer from the excited state luminophore to benzoquinone; or in the case of phenols, hydroquinones and catechols, a benzoquinone deriv. formed directly at the electrode. We have shown that the efficiency of ECL quenching is directly related to both the position of the substituent on the arom. ring and its identity. Results to date as well as potential applications will be presented.

L16 ANSWER 9 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

1998:302409 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 129:73299

Primary cathodic steps of electrogenerated TITLE:

chemiluminescence of lanthanide(III) chelates at oxide-covered aluminum electrodes in aqueous solution Kulmala, S.; Kulmala, A.; Ala-Kleme, T.; Pihlaja, J.

CORPORATE SOURCE: Department of Chemistry, University of Turku, Turku,

FIN-20014, Finland Analytica Chimica Acta (1998), 367(1-3), 17-31 SOURCE:

CODEN: ACACAM; ISSN: 0003-2670

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal English LANGUAGE:



AB The primary processes occurring at cathodically polarized oxide-covered aluminum electrode are discussed in detail. It is pointed out that more energetic cathodic processes can be induced in aq. media at thin insulating film-coated electrodes than at any semiconductor or active metal electrode. It is proposed that tunnel emission of hot electrons with energies well above the level of the conduction band edge of water occur, and the thermalization and solvation of the emitted electrons can result in generation of hydrated electrons. The cathodically pulse-polarized oxide-covered aluminum also generates a strong oxidant (or oxidants) at the oxide/electrolyte interface, and it is proposed that this species is the hydroxyl radical which is generated either by cathodic high field-induced ejection of self-trapped holes as oxygen diamions (i.e. oxide radical ions) into the electrolyte soln., or by the action of anion vacancies and/or F+-centers as the primary oxidants capable of oxidizing hydroxide ions or the hydroxyl groups of the hydroxylated surface on the oxide film. These radicals, hydrated electrons/hydroxyl radicals, can act as mediating reductants/oxidants in redn./oxidn. of solutes. The formation of the primary species is monitored by

electrochemiluminophores which cannot be cathodically excited at active metal electrodes in fully aq. solns., but which can be chem. excited in aq. media in the simultaneous presence of highly reducing and highly oxidizing radicals.

REFERENCE COUNT:

THERE ARE 39 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER (10 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER:

1997:486258 HCAPLUS

TITLE:

SOURCE:

Fluorescent pressure sensitive coatings without paint

binders.

AUTHOR (S):

West, Jeremy P.; Oglesby, Donald M.; Brown, Kenneth;

Upchurch, Billy T.

CORPORATE SOURCE:

Langley Research Center, NASA, Hampton, VA, 23681, USA Book of Abstracts, 214th ACS National Meeting, Las Vegas, NV, September 7-11 (1997), CHED-085. American

Chemical Society: Washington, D. C.

CODEN: 64RNAO

DOCUMENT TYPE:

Conference; Meeting Abstract

LANGUAGE: English

In the past five years the use of coatings contg. luminescent mols. which are quenched by oxygen has become an important tool for obtaining global pressure measurements on model surfaces in wind tunnel testing. Virtually all of these coatings are based on mixing the luminophore with some sort of oxygen permeable polymer which serves as a binder for the coating. The mixt. is applied to the test model surface like a paint, thus the name pressure sensitive paint (PSP) is used to describe such coatings. Some high speed flight research requires testing at temps. as low as -150.degree.C. At these temps. binders become impermeable to oxygen. By chem. or electrochem. modifying the model metal surface we have produced oxygen pressure sensitive coatings which respond to very low (less than 500 ppm) oxygen concns. at temps. as low as -150.degree.C.

L16 ANSWER 11 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1997:346015 HCAPLUS

DOCUMENT NUMBER:

127:101493

TITLE:

SOURCE:

Generation of free radicals and

electrochemiluminescence at pulse-polarized

oxide-covered silicon electrodes in aqueous solutions Ala-Kleme, Timo; Kulmala, Sakari; Latva, Martti

Department of Chemistry, University of Turku, Turku,

FIN-20014, Finland

CORPORATE SOURCE:

AUTHOR(S):

Acta Chemica Scandinavica (1997), 51(5), 541-546

CODEN: ACHSE7; ISSN: 0904-213X

PUBLISHER: Munksgaard DOCUMENT TYPE: Journal LANGUAGE: English

Cathodic pulse-polarization of thin-oxide-film-covered highly doped silicon electrodes induces tunnel emission of hot electrons into aq. electrolyte solns. probably resulting in an **electrochem**. generation of hydrated electrons. Generation of hydrated electrons allows simultaneous prodn. of sulfate radicals from peroxydisulfate ions, and hence, highly reactive radicals are generated in the vicinity of the electrode surface. Generated primary radical species can induce strong redox luminescence from various org. chemiluminophores and luminescent metal chelates, e.g., some lanthanide and transition metal chelates can be detected below nanomolar levels with a linear range of calibration curves of over six orders of magnitude.

L16 ANSWER (12) OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1997:107484 HCAPLUS

DOCUMENT NUMBER: 126:115394

Simultaneous assay method using lanthanide chelates as TITLE:

the luminophore for multiple labels

INVENTOR(S): Billadeau, Mark A.; Leland, Jonathan K.; Shen, Lihong;

Gudibande, Satyanarayana R.

Igen, Inc., USA PATENT ASSIGNEE(S):

SOURCE: PCT Int. Appl., 35 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE \_\_\_\_\_ 19960606 WO 9641177 19961219 WO 1996-US9870 Α1 W: AL, AM, AT, AU, AZ, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG RW: KE, LS, MW, SD, SZ, UG, AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN 19961230 AU 1996-64766 19960606 AU 9664766 A1 PRIORITY APPLN. INFO .: 19950607 US 1995-485715 WO 1996-US9870 19960606

An electrochemiluminescent (ECL) detection method for simultaneous assays is disclosed. The method employs an excitation process whereby .gtoreq.1  ${\tt luminophores}$  are excited through an antenna mechanism by using an oxidative or reductive coreactant. luminophores can be lanthanide chelates, and a persulfate system is used to generate ECL. The ECL emissions may be sepd. either by measuring at different emission wavelengths or by electrode potentials.

```
L16 ANSWER 13 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1997:100066 HCAPLUS
```

126:220025 DOCUMENT NUMBER:

"Reagentless" flow analysis determination of hydrogen TITLE: peroxide by electrocatalyzed luminol chemiluminescence

Atwater, James E.; Akse, James R.; Dehart, Jeffrey; AUTHOR(S):

Wheeler, Richard R., Jr.

UMPQUA Research Company, Myrtle Creek, OR, 97457, USA CORPORATE SOURCE:

Analytical Letters (1997), 30(1), 21-31 SOURCE:

CODEN: ANALBP; ISSN: 0003-2719

PUBLISHER: Dekker Journal DOCUMENT TYPE: LANGUAGE: English

Flow anal. methodol. is presented for the detn. of micromolar levels of aq. H202 by luminol chemiluminescence. Adjustment of the pH to alk. conditions and addn. of the luminophore are implemented using in-line flow-through beds contg. cryst. media. The requirement for a homogeneous catalyst is eliminated by electrochem. initiation of luminescence. Traditional lab. reagents are replaced by packed beds contg. solid phase basification media, and cryst. luminol. The methodol. is suitable for application in both continuous flow and flow injection anal. configurations.

L16 ANSWER(14)OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

1991:459324 HCAPLUS ACCESSION NUMBER:

115:59324 DOCUMENT NUMBER:

TITLE: Electrochemical behavior of polycyclic arenes - activators of peroxide-oxalate

chemiluminescence. The oxidation potential of substituted naphthalenes and anthracenes as their

efficiency criterion

Antonkina, O. A.; Smirnov, S. K.; Gitel', P. O.; AUTHOR(S):

Kryukova, E. V.; Tomilov, A. P.

Gos. Soyuzn. Nauchno-Issled. Inst. Org. Khim. CORPORATE SOURCE:

Tekhnol., Moscow, USSR

Elektrokhimiya (1991), 27(3), 394-401 SOURCE:

CODEN: ELKKAX; ISSN: 0424-8570

DOCUMENT TYPE: Journal LANGUAGE: Russian

Studies of effectiveness of substituted naphthalenes and anthracenes were carried out for the activation of peroxide-oxalate chemiluminescence. The equation correlating the energetic characteristics of chemiluminescence and the oxidn. potential is discussed. An anal. of parameters of the derived equation was performed.

L16 ANSWER 15 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1977:143370 HCAPLUS

DOCUMENT NUMBER: 86:143370

Aircraft probing of atmospheric ozone using a TITLE:

chemiluminescence method

Osechkin, V. V.; Gushchin, G. P.; Pribytkov, L. D.; AUTHOR(S):

Samoilovich, V. G.; Dmitriev, M. T.

CORPORATE SOURCE:

Trudy Glavnoi Geofizicheskoi Observatorii im. A.I. SOURCE:

Voeikova (1976), 357, 161-9 CODEN: TGGOAJ; ISSN: 0376-1274

DOCUMENT TYPE: Journal

Russian LANGUAGE: An aircraft ozonometer with a sensing element using a

chemiluminophore prepd. from rhodamine S [12627-64-4], based on the technol. of V. H. Regener (1964) with slight change, is described. Aircraft sounding of atm. 03 is done by chemiluminescent anal. of air withdrawn by compressor from the engine of a reactive aircraft in flight in the upper troposphere and lower stratosphere. Comparison of a series of parallel 03 detns. made by chemiluminescent and electrochem. ozonometers showed that the chemiluminescence method had greater sensitivity. A trace of atm. O3 concn. (vs. time) detd. by chemiluminescent ozonometer is presented for an April 1973 Moscow to Tashkent flight. The chemiluminescent ozonometer can be used to make stationary as well as in-flight measurements.

L16 ANSWER 16 OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN

1976:515398 HCAPLUS ACCESSION NUMBER:

85:115398 DOCUMENT NUMBER:

Nondestructive method for determining the reliability TITLE:

of electroluminescent indicators



```
AUTHOR(S):
                          Nekrasov, V. M.; Dadeko, L. I.
CORPORATE SOURCE:
                          Kiev. Politekh. Inst., Kiev, USSR
                          Dielektriki i Poluprovodniki (1973), 3, 113-18
SOURCE:
                          CODEN: DLPLAG; ISSN: 0136-3328
DOCUMENT TYPE:
                          Journal
                          Russian
LANGUAGE:
    A method is described for the nondestructive testing of
     electroluminescent sign indicators, based on the study of the
     electrochem. aging processes of ZnS-Cu luminophors. The
     method allows detn. of the av., min., and max. lifetimes of the indicators
      from the measured values of the rectified c.d.,
L16 ANSWER (17) OF 17 HCAPLUS COPYRIGHT 2003 ACS on STN
ACCESSION NUMBER:
                          1975:555129 HCAPLUS
DOCUMENT NUMBER:
                          83:155129
TITLE:
                          Effect of surface-modifying coatings on the stability
                          of powdered electroluminophors
                          Soshchin, N. P.; Talviste, E.; Tammik, A.
AUTHOR(S):
CORPORATE SOURCE:
                          Mater. Vses. Soveshch. Elektrolyumin., 5th (1974),
SOURCE:
                          Meeting Date 1973, 42-7. Editor(s): Fok, M. V.;
                          Rybalkina, L. P. Vses. Nauchno-Issled. Inst.
                          Lyuminoforov Osobo Chist. Veshchestv: Stavropol, USSR.
                          CODEN: 31IVAH
                          Conference
DOCUMENT TYPE:
LANGUAGE:
                          Russian
     Study of the aging of electroluminescent phosphors coated with
     silicate (Sr2SiO4), siloxane, phosphate (NaiO.nP2O5), ZnS, CdS, and NaOH
      (10-3 wt. % soln.) showed that the change in luminescence
     brightness with time had 2 stages for each sample, indicating the presence
     of aging centers with a high degrdn. rate in addn. to electrochem
      . aging centers on the phosphor surface. All of the coatings except NaOH
     soln. decreased the aging rates in both stages. A ref. sample lost 46% of
     its brightness after 350 hr, whereas a CdS-coated sample lost only 22%.
=>
=>
=> select hitrn 115 1-31
'HITRN' IS NOT A VALID FIELD CODE FOR FILE 'HCAPLUS'
ENTER DISPLAY CODE (TI) OR ?:end
=> select hit rn 115 1-31
NO E#s ASSIGNED
=> select hit rn 116 1-17
NO E#s ASSIGNED
=> d stat que
               1 SEA FILE=REGISTRY ABB=ON PLU=ON ELECTROCHEM/BI
T.1
              13 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINOPHOR/BI
L2
              1 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINOPHORE/BI
L3
          13142 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINESC?
L7
         275185 SEA FILE=HCAPLUS ABB=ON PLU=ON L1 OR ?ELECTROCHEM? 230077 SEA FILE=HCAPLUS ABB=ON PLU=ON L7 OR ?LUMINESC?
1.8
L9
            4251 SEA FILE=HCAPLUS ABB=ON PLU=ON L8(L)L9
L10
                                                  2 TERMS
L11
                SEL PLU=ON L3 1- CHEM:
               1 SEA FILE=HCAPLUS ABB=ON PLU=ON L11
L12
            3990 SEA FILE=HCAPLUS ABB=ON PLU=ON L12 OR L2 OR ?LUMINOPHOR?
L13
                                          PLU=ON
                                                  L10 AND L13
              48 SEA FILE=HCAPLUS ABB=ON
T.14
              31 SEA FILE=HCAPLUS ABB=ON PLU=ON L14 AND (REACTION OR COURSE
```

OR BINDING)



=> d ibib abs hitrn 119 1-8

L19 ANSWER (1) OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2003:326945 HCAPLUS

DOCUMENT NUMBER: 139:62064

TITLE: Coordination Properties of a Diarylaza Crown Ether

Appended with a Luminescent [Ru(bipy)3]2+

Un.

AUTHOR(S): Charbonniere, Loiec J.; Ziessel, Raymond F.; Sams,

Craig A.; Harriman, Anthony

CORPORATE SOURCE: Laboratoire de Chimie Moleculaire UMR 7008 au CNRS,

Ecole de Chimie Polymere et Materiaux, Strasbourg,

67087, Fr.

SOURCE: Inorganic Chemistry (2003), 42(11), 3466-3474

CODEN: INOCAJ; ISSN: 0020-1669

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

LANGUAGE:

Me O O O

The [Ru(bipy)2(1)](PF6)2 (bipy refers to 2,2'-bipyridine) complex, comprising a Ru(II) tris(2,2'-bipyridine) luminophor covalently linked to a di[(o-triethyleneglycoxy)phenyl]amine crown ether (1 = I), was synthesized and fully characterized. The photophys. properties of this metal complex were examd. in soln. at ambient temp. Luminescence from the metal complex is enhanced significantly in the presence of various adventitious cations, including protons. In particular, Li+cations bind to the crown ether, as evidenced by 1H NMR and luminescence spectroscopy. Cation binding serves to decrease the rate of reductive quenching of the triplet state of the metal complex, thereby increasing the extent of luminescence. The soln.-phase conformation of [Ru(bipy)2(1)](PF6)2, with and without encapsulated Li+, was examd. by 2-dimensional NMR and by mol. dynamics simulations.

Т

was examd. by 2-dimensional NMR and by mol. dynamics simulations.

REFERENCE COUNT: 41 THERE ARE 41 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L19 ANSWER 2 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 2002:834791 HCAPLUS DOCUMENT NUMBER: 137:289880



TITLE: Composition and methods for high-density biochip

preparation for therapeutic and diagnostic application

INVENTOR(S): Guo, Zhanjun; Zhao, Hua; Guo, Aiqin

PATENT ASSIGNEE(S): Peop. Rep. China

SOURCE: Faming Zhuanli Shenqing Gongkai Shuomingshu, 18 pp.

CODEN: CNXXEV

DOCUMENT TYPE: Patent Chinese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE ·
PF	CN 1333376 RIORITY APPLN. INFO		20020130	CN 2001-121535 CN 2001-121535	
AE	The biochip combundle of org. plastic optical probe, cDNA promedicine recept on or in the firmicro-groove, rand region between the polymeric fiber metal wire, etc.) deposited in labeling reager process comprisactivated capital witting with microstructure, The high-d. bio	nsists of or inormal fiber, pobe, PNA tor, poliber carmicro-slaween the lary fiber, glasses; which microsses fixing the laber, patter, pothip ma	g. polymeric and/or meta probe, mRNA ysaccharide rier; microtic, microturer (such as fiber, optoblem are coated ructure, patabilized wing different carriers ling capillan, and regicy be used in	fiber carrier units of fiber, glass fiber, al wire, etc.); bio-a d, polypeptide, antig agglutinin, cell, an atructure (such as minube, and/or micro-mester units; and labeli a or a bundle of orgical fiber, plastic of with colored dye or tern, and region. The thing cose (at a ratic kinds of bio-macrom ary fibers or overlap on; and depositing the gene sequencing, dind research of life s	(such as a or a optical fiber, rray (such as DNA en, antibody, d/or tissue) fixed cro-pits, h, etc.), pattern, ng reagent and or inorg. ptical fiber, and/or luminophor he bio-array and of 1:10-30). The ols. on or in the ertion methods; ping; forming e labeling reagent. agnosis of disease,
		_			

L19 ANSWER (3) OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2002:578286 HCAPLUS

DOCUMENT NUMBER: 138:92129

TITLE: Adsorptive pressure-sensitive coatings for unsteady

flow measurements

AUTHOR(S): Teduka, Norikazu; Kameda, Masaharu; Asai, Keisuke;

Amao, Yutaka; Nakakita, Kazuyuki

CORPORATE SOURCE: Dept. of Mechanical Systems Eng., Tokyo University of

Agriculture and Technology, Koganei-shi, Tokyo,

184-8588, Japan

SOURCE: Nippon Kikai Gakkai Ronbunshu, B-hen (2002), 68(669),

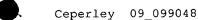
1391-1399

CODEN: NKGBDD; ISSN: 0387-5016

PUBLISHER: Nippon Kikai Gakkai

DOCUMENT TYPE: Journal LANGUAGE: Japanese

AB Time response of pressure-sensitive luminescent coatings has been investigated theor. and exptl. The present coating consists of a thin anodized Al layer formed by electro-chem. process and luminophores adsorbed by chem. or phys. adsorption. The theor. anal. shows that the effective diffusion coeff. for O permeation in the anodized Al layer is up to 1 .times. 10-5 m2/s, which implies that the present coating should have the response time of the order of microseconds. Two porphyrin compds. and a ruthenium(II) complex were used as the luminophores and the response to a step change in pressure was studied using a pressure jump app. and a shock tube. The response time of coating with tris(4,7-diphenylphenanthroline)ruthenium(II) was >20 .mu.s and showed dependence on the anodized Al thickness, while



the response time for tetrakis-(4-carboxyphenyl)porphyrin coating was <10 .mu.s and showed independent of the Al thickness.

L19 ANSWER 4 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2002:123352 HCAPLUS

DOCUMENT NUMBER:

136:163679

TITLE:

Capsules encapsulating solid particles of

signal-generating organic substances and their use in in vitro bioassays for detection of target molecules

in a sample

INVENTOR(S):

Trau, Dieter; Renneberg, Reinhard; Caruso, Frank;

Lehmann, Matthias

PATENT ASSIGNEE(S):

8 Sens. Biognostic A.-G., Germany

SOURCE:

PCT Int. Appl., 49 pp.

DOCUMENT TYPE:

CODEN: PIXXD2

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PA:	PATENT NO.				PATENT NO. KIND			D	DATE							ο.	DATE			
	2002								WO 2001-EP9114 200108						0807					
WO		ΑE,	AG,	AL,	AM,	AT,	ΑU,							BZ,						
														GD, LC,						
														ΝΖ, UA,						
		VN,	YU,	ZA,	ZW,	AM,	ΑZ,	BY,	KG,	ΚZ,	MD,	RU,	ТJ,	TM						
	RW:													AT, PT,						
DE	1004	вJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,					
DE	1004	2023		C2 20030410					DE 2000-10042023 2000080											
AU EP	AU 2001091746 EP 1309867				5 2	20020218			A) El	U 200 P 200	01-9: 01-9:	1746 7188	4	20010	0807 0807					
		ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,			NL,		MC,	PT,			
PRIORIT	Y APP	•	•		ъ∨,	FI,	KU,		DE 2	000-	1004			20000						

AB The present invention refers to capsules encapsulating solid particles of signal-generating org. substances and carrying on the outer surface affinity mols. for specific recognition of and binding to target mols. in a sample. The invention is directed to the use of these capsules for signal prodn. in the optical, electrochem. or chem. detection of target mols. To obtain a signal the signal-generating org. substances are released and dissolved. A detection method and a kit are provided. Polyelectrolyte capsules encapsulating solid fluorescein diacetate particles were prepd. from poly(allylamine hydrochloride) and poly(sodium 4-styrenesulfonate) or alginic acid. Antibody was coupled to the outer surface of the capsules for use in immunoassays.

L19 ANSWER 5 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NOMBER:

1993:178760 HCAPLUS

DOCUMENT NUMBER:

118:178760

TITLE:

Spectroelectrochemical characteristics and

photophysics of a series of ruthenium(II) complexes with 1,4,5,8,9,12-hexaazatriphenylene: effects of

polycomplexation

AUTHOR(S): Jacque CORPORATE SOURCE: Univ.

Jacquet, Luc; Kirsch-de Mesmaeker, Andree Univ. Libre Bruxelles, Brussels, 1050, Belg. Journal of the Chemical Society, Faraday Transactions

SOURCE:



(1992), 88(17), 2471-80 CODEN: JCFTEV; ISSN: 0956-5000

DOCUMENT TYPE: Journal LANGUAGE: English

A study of the photophys. mechanisms controlling the luminescence lifetimes of a series of Ru2+ complexes based on the same [Ru3+-hat.bul.-]\* luminophore (hat = 1,4,5,8,9,12hexaazatriphenylene), and extending from monometallic building blocks to polynuclear compds., is presented, together with the relevant spectroscopic and redox data. The parameters which characterize these different mechanisms, i.e. the radiative and nonradiative deactivation rate consts. of the 3MLCT (metal to ligand charge transfer) state and the thermal activation from this state to the 3MC (metal centered) and/or to another 3MLCT state, have been detd. from the analyses of the curves obtained by plotting the lifetimes or intensities of luminescence as a function of temp. The math. and exptl. limitations of the anal. method have been investigated and detn. of the parameters has been improved by using a new global anal. algorithm. Interestingly the comparison of the data for the monometallic and polynuclear complexes indicates that the (Ru3+-hat.bul.-)\* luminophore, which is common for the whole series, presents characteristics that change when the hat becomes a bridging ligand, i.e. those that depend on the vibration modes of the complex. Moreover, for these polynuclear compds., there is

no contribution of the 3MC state to the mechanism of control of the

L19 ANSWER 6 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1990:583676 HCAPLUS

DOCUMENT NUMBER: 113:183676

luminescence lifetime.

TITLE: Dual luminescence from a mixed-metal complex

containing rhenium(I) and ruthenium(II)

photochromophores

AUTHOR(S): Van Wallendael, Shawn; Rillema, D. Paul

CORPORATE SOURCE: Dep. Chem., Univ. North Carolina, Charlotte, NC,

28223, USA

Journal of the Chemical Society, Chemical SOURCE:

Communications (1990), (16), 1081-2 CODEN: JCCCAT; ISSN: 0022-4936

Journal DOCUMENT TYPE: LANGUAGE: English

[(Bpy)2Ru(bpyen)Re(CO) $\bar{3}$ (py)(PF6)3 [bpy = 2,2'-bipyridine; bpyen = 1,2-bis(4'-methyl-2,2'-bipyridyl-4-yl)ethane] was prepd. by reaction of Re(bpyen)(CO)3Cl with [Ru(bpy)2(MeOH)](PF6)2 followed by AgPF6 and pyridine. The cation luminesces at 540 and 610 nm in CH2Cl2 at room temp. The luminescence at 540 nm is derived from the Re luminophore component and that at 610 nm from the Ru luminophore component.

L19 ANSWER 7 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1982:134917 HCAPLUS

DOCUMENT NUMBER: 96:134917

TITLE: Amperometric determination of lanthanum using

N-(m-nitrocinnamoyl) phenylhydroxylamine

Oliferenko, G. L.; Gallai, Z. A.; Sheina, N. M.; AUTHOR(S):

Svediene, N.

CORPORATE SOURCE: M. V. Lomonosov Moscow State Univ., Moscow, USSR SOURCE:

Zhurnal Analiticheskoi Khimii (1981), 36(12), 2330-4

CODEN: ZAKHA8; ISSN: 0044-4502

DOCUMENT TYPE: Journal Russian LANGUAGE:

N-(m-Nitrocinnamoyl)phenylhydroxylamine (I), a new reagent in the class of unsatd. hydroxamic acids, was synthesized; its pK is 7.7 .+-. 0.03, and its electrochem. oxidn. at a rotating graphite microelectrode

was investigated. I reacts with La to form a slightly sol. compd. with a metal:reagent ratio of 1:3 at pH 7.1. The compd. soly. under titrn. conditions is 9 .times. 10-6 M. The La(III) detection limit is 4 .mu.g/mL. Amperometric detn. of La in  $\operatorname{luminophors}$  with the formula La2BaOxSy.Eu (3-5%) was demonstrated.

L19 ANSWER 8 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1967:109930 HCAPLUS

DOCUMENT NUMBER: 66:109930

Light output of powdered electroluminophors TITLE:

and its dependence on the grain size composition

AUTHOR(S):

Chukova, Yu. P. Optika i Spektroskopiya (1967), 22(3), 450-5 SOURCE:

CODEN: OPSPAM; ISSN: 0030-4034

DOCUMENT TYPE: Journal LANGUAGE: Russian

The energy absorption and light output of ZnS luminophors are examd. An empirical dependence of the light output on the external potential yields data that agree well with expts. for a wide range of potentials (50-1000 v.). The dependence of the absorbed energy on the effective size of the ZnS grains is investigated for 3 types of distribution of the absorbing barriers: const. barrier d. per vol., per surface, and fixed no. of barriers per particle. The calcd. data are compared with exptl. (Lehmann, **Electrochem**. Soc. 105, 585(1958)), revealing that the barrier no. is const. for ZnS crystals of similar sizes. The character of the distribution of energy-absorbing barriers is markedly analogous to that of the distribution of radiating barriers.

```
=> d stat que
             13 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINOPHOR/BI
L2
              1 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINOPHORE/BI
L3
                                                2 TERMS
                SEL PLU=ON L3 1- CHEM:
L11
              1 SEA FILE=HCAPLUS ABB=ON PLU=ON L11
1.12
           3990 SEA FILE=HCAPLUS ABB=ON PLU=ON L12 OR L2 OR ?LUMINOPHOR?
L13
              8 SEA FILE=HCAPLUS ABB=ON PLU=ON L13 AND ((BIND? OR RATE)(2A)RE
L21
                ACTION OR TIME (W) COURSE)
=>
=> d ibib abs hitrn 121 1-8
L21 ANSWER OF 8
ACCESSION NUMBER:
                   HCAPLUS COPYRIGHT 2003 ACS on STN
                         2003:439004 HCAPLUS
DOCUMENT NUMBER:
                         139:164872
                         Syntheses and Characterization of Upper Rim 1,2- and
TITLE:
                         1,3-Diphosphinated Calix[4] arenes and Their
                         Corresponding 1,5-Cyclooctadienylrhodium(I) Complexes:
                         Comparison of the Catalytic Hydroformylation
                         Properties of Terminal Alkenes
                         Plourde, Francois; Gilbert, Karine; Gagnon, Jonathan;
AUTHOR(S):
                         Harvey, Pierre D.
                         Departement de Chimie, Universite de Sherbrooke,
CORPORATE SOURCE:
                         Sherbrooke, QC, J1K 2R1, Can.
                         Organometallics (2003), 22(14), 2862-2875
SOURCE:
                         CODEN: ORGND7; ISSN: 0276-7333
                         American Chemical Society
PUBLISHER:
DOCUMENT TYPE:
                         Journal
                         English
LANGUAGE:
    Rhodium and iridium dimeric cyclooctadiene complexes with calix[4]arene
    upper rim diphosphines were prepd., characterized and tested for
    hydroformylation activity. The 5,11- (9a,b) and 5,17-
    bis(dialkylphosphino)-25,26,27,28-tetra-n-propoxycalix[4]arene (5a,b)
     ligands (a, alkyl = Me; b, iPr) were prepd. and coordinated to Rh(COD) +
     fragments (COD = 1,5-cyclooctadiene). The ligands 5,17-
    bis(diphenylphosphino)-11,23-dibromo-25,26,27,28-tetra-n-
    propoxycalix[4]arene (5c) and 5,11-bis(diphenylphosphino)-25,26,27,28-
     tetra-n-propoxycalix[4] arene (9c) were coordinated to M(COD)+(M=Rh, Ir)
     and RhCl(CO) fragments, as well. On the basis of mass spectrometry and
     31P NMR spin-lattice relaxation time measurements (T1), all of the
     complexes are found to be dimers. Mol. modeling provides evidence that
     ring stress favors the dimer over the monomer, and the modeled structures
     for both 5,11- and 5,17-isomers were corroborated by the comparison of the
    photophysics of the [5c.cntdot.Ir(COD)][PF6]2 (14) species at 77 K.
     decrease in emission lifetimes of the P2Ir(C:C)2+ luminophor in
     the presence of 1-hexene is more pronounced for the 5,17-isomer,
     indicating reduced steric hindrance about the metallic center. The
    catalytic hydroformylation of 1-hexene, styrene and vinyl esters alkenes
     using [LRh(COD)]22+ catalyst precursors (L = 5a-c, 9a-c) was tested under
     various conditions. The basicity and the cone angle of the phosphines had
     an impact on the n/iso ratios of the resulting aldehydes. The turnover
     frequencies (tof's) are generally larger for the 5,11-series for 1-hexene,
    but depend on the nature of the phosphine for styrene, vinyl acetate,
     vinyl benzoate, and vinyl p-tert-butylbenzoate.
                               THERE ARE 46 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                         46
                               RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
```

L21 ANSWER 2 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NOWBER: 2000:268364 HCAPLUS

DOCUMENT NUMBER: 133:55520

~)

Indole-3-pyruvic and -propionic acids, kynurenic acid, TITLE:

and related metabolites as luminophors and

free-radical scavengers

AUTHOR(S): Hardeland, R.; Zsizsik, B. K.; Poeggeler, B.;

Fuhrberg, B.; Holst, S.; Coto-Montes, A.

Institut fur Zoologie und Anthropologie, Universitat CORPORATE SOURCE:

Gottingen, Gottingen, D-37073, Germany

Advances in Experimental Medicine and Biology (1999), SOURCE:

467 (Tryptophan, Serotonin and Melatonin), 389-395 CODEN: AEMBAP; ISSN: 0065-2598

PUBLISHER: Kluwer Academic/Plenum Publishers

DOCUMENT TYPE: Journal English LANGUAGE:

Chemiluminescence assocd. with oxidn. by free radicals was investigated in an alk., hemin-catalyzed hydrogen peroxide system, using the following tryptophan metabolites as radical scavengers: indole-3-pyruvic, indole-3-propionic, kynurenic, xanthurenic and quinaldic acids and 4-hydroxy-quinoline. Light emission from oxidn. of the indolic compds. was only partially inhibited by the hydroxyl-radical scavenger DMSO, but strongly suppressed by the superoxide-anion scavenger Tiron, whereas chemiluminescence generated from kynurenic acid was strongly inhibited by either of these compds. Light emission from oxidn. of kynurenic acid lasts for a surprisingly long period of time: At 0.4 mM and 20.degree.C, luminescence increased for 5 h and continued at a high rate for more than a day. Comparison of structural analogs indicated that the 4-hydroxyl and carboxyl groups of kynurenic acid are essential for effective light emission, and that an addnl. 8-hydroxyl residue leading to an intramol. hydrogen bond diminishes the reaction rate.

THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 14 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L21 ANSWER (3 ) F 8 HCAPLUS COPYRIGHT 2003 ACS on STN

1996:464529 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 125:109657

Rate measurements of biomolecular reactions using TITLE:

electrochemiluminescence

this opplication Nacamulli, Laurette; Leland, Jonathan K.; Hayes, INVENTOR(S):

Stephanie A. Igen, Inc., USA

PATENT ASSIGNEE(S):

PCT Int. Appl., 55 pp. SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE: Patent English LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.					KIND DATE					APPLICATION NO. DATE							
WO	WO 9617248			A1 19960606				W									
	W:													DK,			
														LT,			
		MG,	MN,	MW,	MX,	NO,	ΝZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	TJ,
		TM,													~-	-	
														FR,			
		> <del>}.</del> ,.				PT,	SE,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	МĿ,	MR,
		ΝE,	SN,	$\sqrt{TD}$ ,													
US	5527	710		<i>)</i> A		1996	0618		-		94-3		-	1994			
_CA	2206	335_		A	A	1996	0606		C.	A 19	95-2	2063.	35	1995	1204		
AU 9645966		A.	1	1996	0619		A	U 19	96-4	5966		1995	1204				
ΑU	711459 B2				2	1999	1014										
JP 10508104			T	2	1998	0804		J	P 19	95-519172		2	1995	1204			
EΡ	8718	91		A	1	1998	1021		EP 1995-944075 19951						1204		

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE PRIORITY APPLN. INFO.: US 1994-347984 19941202 19951204 WO 1995-US15982

A biomol. reaction, such as an enzymic or affinity binding reaction, is effected in an electrochem. cell with a mixt. of reagents including a luminophor which will relate the concn. of a reaction, a reaction partner, or the reaction product of a reaction partner to the electrochemiluminescent intensity to det. the rate of the biomol. reaction. The reaction partner is a reagent which reacts with the reactant and which itself or its reaction product participates with the luminophor to cause ECL emissions. The ECL intensity is modulated with a series of elec. pulses which are applied to the mixt. of reagents at a preselected potential for preselected intervals of time and duration. The intensity is measured during those intervals to provide a timed series of values (P). The same expt. is repeated twice more, once wherein the modulation is conducted after the reaction has gone to completion to obtain the timed series of values (C), and the last time the reaction is performed in the absence of the reaction partner to obtain the timed series of values (B). The results are normalized (N) using the formula N = (P - B)/(C - B) to obtain a series of values (N) which are plotted to obtain the time course of reaction.

L21 ANSWER 4 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1996:260414 HCAPLUS

DOCUMENT NUMBER: 124:336707

Initiation of degenerate-branched chain reaction of TITLE:

glycine deamination with ultraweak UV irradiation or

hydrogen peroxide

Voeikov, V. L.; Baskakov, I. V.; Kafkalias, K.; AUTHOR(S):

Naletov, V. I.

Moscow State Univ., Biological Fac., Moscow, 119899, CORPORATE SOURCE:

Russia

Bioorganicheskaya Khimiya (1996), 22(1), 39-47 SOURCE:

CODEN: BIKHD7; ISSN: 0132-3423

MAIK Nauka PUBLISHER: DOCUMENT TYPE: Journal LANGUAGE: Russian

Short irradn. of an aq. soln. of glycine by an ultraweak source of UV photons initiates a process accompanied by the accumulation of ammonium ions. More intense deamination occurs in an aq. soln. of glycine in the presence of H2O2. The kinetics of the ammonia formation coincides with the kinetics of enhancement of luminescence, which was registered in the presence of a luminophor, ethidium bromide or tyrosine. When the process is completed, less than 5% glycine is consumed. At the initial step of the reaction, the concn. of H2O2 in the soln. increases significantly in comparison with the starting concn. but then begins to decrease. If the luminophor concn. exceeds the optimum, the reaction rate enhancement decelerates and the ammonia yield decreases. UV photons and H2O2 are assumed to initiate the degenerate-branched chain reaction of glycine deamination.

L21 ANSWER  $\int \Phi F 8$ HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1994:252237 HCAPLUS

DOCUMENT NUMBER: 120:252237

TITLE: Photoassisted oxidation of formaldehyde in

luminophore-grade CdS-Pt suspensions

Bamwenda, Gratian; Mika, Alicja M.; Winnicki, Tomasz AUTHOR(S):

Inst. Environ. Prot. Eng., Tech. Univ. Wroclaw, CORPORATE SOURCE:

Wroclaw, 50-370, Pol.

Environment Protection Engineering (1992), Volume Date SOURCE:

1990, 16(2), 67-73

CODEN: EPEND9; ISSN: 0324-8828

DOCUMENT TYPE: Journal LANGUAGE: English

Photoassisted oxidn. of HCHO in aq. solns. with a simultaneous evolution of mol. H was studied in platinized luminophore-grade CdS suspensions. The course of the reaction depended on illumination time, amt. of Pt deposited onto CdS surface, and HCHO soln. pH. The apparent first-order reaction rate const. was 1.48 .times.  $10-4/\min$ .

L21 ANSWER 6 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1981:626405 HCAPLUS

95:226405 DOCUMENT NUMBER:

Effect of zinc oxide nonstoichiometry on the kinetics TITLE:

of the formation of a manganese-activated zinc

silicate (Zn2SiO4:Mn) luminophor

Korablev, N. M.; Gavrilov, V. V.; Semin, E. G.; AUTHOR(S):

Vagina, T. Sh.; Vinnikova, V. I.

CORPORATE SOURCE: USSR

Sbornik Nauchnykh Trudov - Vsesoyuznyi SOURCE:

Nauchno-Issledovatel'skii Institut Lyuminoforov i

Osobo Chistykh Veshchestv (1979), 18, 45-9

CODEN: SNVNAR; ISSN: 0371-1722

DOCUMENT TYPE: Journal Russian LANGUAGE:

The effect of  ${\mbox{ZnO}}$  nonstoichiometry on the rate of formation of a  ${\mbox{Zn2SiO4}}$ phosphor by reaction of ZnO with SiO2 in the presence of a Mn activator was studied x-ray diffractometry, luminescence spectroscopy and ESR. The rate of reaction and its completeness increase as the no. of vacancies in the initial ZnO increases. The limiting reaction step is the formation of a (Zn, Mn) 2SiO4 solid soln.

L21 ANSWER (7) OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

1978:431496 HCAPLUS ACCESSION NUMBER:

89:31496 DOCUMENT NUMBER:

Surface reduction of zinc sulfide and oxide by atomic TITLE:

hydrogen (luminescence study)

Styrov, V. V.; Izmailov, Sh. L.; Sokolov, V. A. AUTHOR(S):

Tomsk. Inst. Avtom. Sist. Upr. Radioelektron., Tomsk, CORPORATE SOURCE:

USSR

Tezisy Dokl. - Soveshch. Kinet. Mekh. Khim. Reakts. SOURCE:

Tverd. Tele, 7th (1977), Volume 3, 99-102. Editor(s): Lyakhov, N. Z. Akad. Nauk SSSR, Sib. Otd., Inst. Fiz.-Khim. Osn. Pererab. Miner. Syr'ya: Novosibirsk,

USSR.

CODEN: 37UOAU

Conference

DOCUMENT TYPE: LANGUAGE: Russian

A new sensitive method for detg. the kinetics of reactions at the gas-solid interface was proposed and tested on the redn. of ZnS (with luminophores Cu, Co, Cl, Mn) and ZnO with at. H. It is based on the detn. of radical-recombination luminescence spectra of the surface during its interaction with at. gases. The method is also suitable for studying lattice defects of the surface.

L21 ANSWER 8 OF 8 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1977:497776 HCAPLUS

DOCUMENT NUMBER: 87:97776

Modeling of neural mechanisms in the TITLE:

electrochemiluminescent cells. Communication I

Leonov, V. P. AUTHOR(S):

CORPORATE SOURCE: USSR

Problemy Bioniki (1976), 16, 99-104 SOURCE:

CODEN: PBNKAV; ISSN: 0555-2656

DOCUMENT TYPE: LANGUAGE:

Journal Russian

Electrochemiluminescent cells (electrochem. cells contg. luminophores in their electrolyte solns.) were analyzed as possible models for synaptic excitation, inhibition, adaptation, and summation. In electrochem. cells contg. 1,5-diphenyl-3-styrylpyrazoline (I), luminescence (synaptic excitation) was obsd. when a I-cation radical reacted with a I-anion radical to form a luminescent excited mol. or when a 2nd electron was transferred to the I-cation radical. The anodic reaction representing the presynaptic input can be used to model summation (facilitation) with the rate of the reaction modeling spatial summation and the life time of the I-cation radical modeling temporal summation. Control of the rate of the cathodic reaction mimics adaptation. In electrochem. cells contg. Na tetraphenylborate plus triphenylimidazole, 1,5-diphenyl-3-(p-chlorophenyl)pyrazoline, or dibenzyl, electrochemiluminescence was obsd. as the result of the formation of exciplexes between the tetraphenylborate anion and the cation radicals of the other compds. This system can mimic the same synaptic mechanism as the system contg. I. In addn., if a neg. impulse is applied to the synaptic electrode of this 2nd system before the pos. impulse, luminescence is inhibited. Thus, synaptic inhibition can also be modeled in this system.

=> =>

=> fil hcaplus FILE 'HCAPLUS' ENTERED AT 17:40:20 ON 08 SEP 2003 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2003 AMERICAN CHEMICAL SOCIETY (ACS)

Copyright of the articles to which records in this database refer is held by the publishers listed in the PUBLISHER (PB) field (available for records published or updated in Chemical Abstracts after December 26, 1996), unless otherwise indicated in the original publications. The CA Lexicon is the copyrighted intellectual property of the the American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited.

FILE COVERS 1907 - 8 Sep 2003 VOL 139 ISS 11 FILE LAST UPDATED: 7 Sep 2003 (20030907/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

```
=>
=>
=> d stat que
              1 SEA FILE=REGISTRY ABB=ON PLU=ON ELECTROCHEM/BI
L1
L2
             13 SEA FILE=REGISTRY ABB=ON
                                          PLU=ON
                                                  LUMINOPHOR/BI
              1 SEA FILE=REGISTRY ABB=ON
                                          PLU=ON
L3
                                                  LUMINOPHORE/BI
          13142 SEA FILE=REGISTRY ABB=ON PLU=ON LUMINESC?
T.7
         275185 SEA FILE=HCAPLUS ABB=ON PLU=ON L1 OR ?ELECTROCHEM?
         230077 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L7 OR ?LUMINESC?
L10
           4251 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L8(L)L9
L11
                SEL PLU=ON L3 1- CHEM:
                                                 2 TERMS
L12
              1 SEA FILE=HCAPLUS ABB=ON
                                          PLU=ON
                                                L11
                                                 L12 OR L2 OR ?LUMINOPHOR?
           3990 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON
L13
                                          PLU=ON
L14
             48 SEA FILE=HCAPLUS ABB=ON
                                                 L10 AND L13
L15
             31 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON
                                                L14 AND (REACTION OR COURSE
                OR BINDING)
L16
             17 SEA FILE=HCAPLUS ABB=ON
                                          PLU=ON L14 NOT L15
                                                 L13 AND ((BIND? OR RATE)(2A)RE
L21
              8 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON
                ACTION OR TIME (W) COURSE)
L22
             44 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON L8 (L) (?MODULAT? (5A) SIGNAL)
             44 SEA FILE=HCAPLUS ABB=ON PLU=ON L22 NOT (L15 OR L16 OR L21)
T<sub>2</sub>3
```

=> d ibib abs hitrn 123 1-44

=> =>

L23 ANSWER 1 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2003:640487 HCAPLUS ACCESSION NUMBER:

TITLE: Spectroelectrochemical sensing: planar waveguides AUTHOR(S): Ross, Susan E.; Shi, Yining; Seliskar, Carl J.;

Heineman, William R.

CORPORATE SOURCE:

Department of Chemistry, University of Cincinnati, PO Box 210172, Cincinnati, OH, 45221-0172, USA

Electrochimica Acta (2003), 48(20-22), 3313-3323 SOURCE:

CODEN: ELCAAV; ISSN: 0013-4686

PUBLISHER: Elsevier Science B.V.

Journal DOCUMENT TYPE: English LANGUAGE:

The spectroelectrochem. sensor combines in a single device electrochem., spectroscopy, and selective partitioning into a film, giving improved selectivity for applications that involve complex samples. Sensing is based on the change in optical signal that accompanies electrochem. modulation of analyte that has partitioned into the film. Two classes of optical quality chem.-selective films based on two different host materials, namely, sol-gel processed silica and cross-linked poly(vinyl alc.) have been developed. Films are typically 400-700 nm thick. Three types of sensor platforms are discussed: a multiple internal reflection (MIR) optic consisting of a bilayer of an indium tin oxide (ITO) optically transparent electrode deposited on a 1-mm thick glass substrate, a planar waveguide in which a potassium ion-exchanged BK7 glass waveguide (5-9 .mu.m thick) was over-coated with a thin film of ITO, and a planar waveguide in which a potassium ion-exchanged BK7 glass waveguide channel was formed and a pair of electrodes deposited along side the channel. These sensors were evaluated with ferrocyanide and a selective film of PDMDAAC-SiO2, where PDMDAAC=poly(di-Me diallylammonium chloride).

L23 ANSWER 2 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NOMBER: 2003:629681 HCAPLUS

TITLE: Studies of the influence of optical effects on spectroelectrochemical sensor's performance

Piruska, Aigars; Zudans, Imants; Seliskar, C. J.; AUTHOR(S):

Heineman, W. R.

CORPORATE SOURCE: Department of Chemistry, University of Cincinnati,

Cincinnati, OH, 45221, USA

Abstracts of Papers, 226th ACS National Meeting, New SOURCE:

York, NY, United States, September 7-11, 2003 (2003), ANYL-019. American Chemical Society: Washington, D.

CODEN: 69EKY9

Conference; Meeting Abstract

DOCUMENT TYPE: LANGUAGE: English

The spectroelectrochem. sensor represents an internal reflection element coated with a thin optically transparent electrode and a chem. selective film. The anal. signal of the sensor is the modulation of light intensity at the chosen wavelength due to reversible electrochem. cycling of the analyte from absorbing to non-absorbing form in to the film. Light propagation in thin films was studied to elucidate influence of purely optical effects on reflectance spectrum. Under certain conditions light interference in film can be satisfied and enhanced absorbance obsd. An absorbance band distortion in the sensors configuration was studied theor. and exptl. Theor. studies were performed using ray optics approach on a model system, that consists of an absorbing film described by a Lorentz oscillator enclosed between semi-infinite, non-absorbing media. Tris(2,2'-bipyridyl) ruthenium(II) chloride uptake in Nafion film have been used as a model system for the exptl. studies.

L23 ANSWER B OF 44 ACCESSION NUMBER: HCAPLUS COPYRIGHT 2003 ACS on STN

2003:466034 HCAPLUS

139:171929 DOCUMENT NUMBER:

TITLE: Photoacoustic and photoelectrochemical current response of nanostructured TiO2 electrodes

AUTHOR(S): Toyoda, Taro; Hayashi, Masashi; Shen, Qing Department of Applied Physics and Chemistry, The

CORPORATE SOURCE: University of Electro-Communications, Tokyo, 182-8585,

Japan

Japanese Journal of Applied Physics, Part 1: Regular SOURCE:

Papers, Short Notes & Review Papers (2003), 42(5B),

3036-3040 CODEN: JAPNDE

Japan Society of Applied Physics PUBLISHER:

Journal DOCUMENT TYPE: English LANGUAGE:

The authors report the effect of voltage in a concd. KCl electrolyte applied to nanostructured TiO2 electrodes during their final prepn. processes on photoacoustic (PA) and photoelectrochem. (PEC) current spectra and their modulation frequency dependence to clarify their response characteristics. The PA signal intensities of the TiO2 electrodes with different applied voltage treatments are higher than that without the treatments below the fundamental absorption edge. These results suggest an inner-band transition due to an increase in carrier concn. owing to donor levels formed by partially reduced Ti ions (Ti4+ .fwdarw. Ti3+) following the applied voltage treatments. The PEC spectra for the applied voltage treatments >-1.5 V show broad bands at .apprx.3.3 and 3.8 eV. The intensities of all peaks increase rapidly at an applied voltage >-1.5 V, which is similar to that of the PA intensity at a photon energy of 2.0 eV. The increase in the PEC intensity with different applied voltage treatments implies an increase in carrier concn. due to donor level formation by the treatments. The modulation frequency dependence of the PA signal intensity is related to the applied voltage treatments below a frequency of 60 Hz. This indicates the increase in the interfacial thermal resistance to prevent heat diffusion at the interface between the nanostructured TiO2 film and the Ti substrate with the increase in applied voltage. The modulation frequency dependence of PEC current also depends on the applied voltage treatments, indicating the enhancement of the photoexcited electron diffusion response with the increase in applied voltage.

THERE ARE 31 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 31 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 4 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2003:389038 HCAPLUS

139:110589 DOCUMENT NUMBER:

Kinetics of oxygen exchange in Sr(Ti0.65Fe0.35)03 TITLE:

Schneider, Th.; Wagner, S. F.; Menesklou, W.; AUTHOR(S):

Ivers-Tiffee, E.

CORPORATE SOURCE: Institut fur Werkstoffe der Elektrotechnik,

Universitat Karlsruhe (TH), Karlsruhe, 76128, Germany

Materials Research Society Symposium Proceedings (2003), 756(Solid State Ionics--2002), 145-150

CODEN: MRSPDH; ISSN: 0272-9172

Materials Research Society PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

SOURCE:

Current limiting electrochem. pumping cells (amperometric sensors) based on zirconia are commonly used for engine control applications. Fast resistive-type sensors adapted from semiconducting metal oxides are a promising alternative for future exhaust gas monitoring systems. Therefore among the interesting characteristics of the materials system Sr(Ti0.65Fe0.35)03, including high sensitivity and temp. independence at high oxygen partial pressures (pO2 > 10-4 bar), a short response time (t90 = 30 ms) is obviously the most salient. The latter is detd. by the kinetics of the oxygen surface transfer and subsequent diffusion of oxygen vacancies V-o. For thin samples and low temps. the surface transfer is dominant, since bulk diffusion usually occurs very fast. The presented model is based on the frequency-domain anal. of amplitude and phase shift of the response signal obtained from a pO2 modulation in a fast kinetic measurement setup. This method allows both the measurement of response times in the sub-millisecond range as well as the distinction of the behavior either controlled by vol.

diffusion or by surface transfer reaction in Sr(Ti0.65Fe0.35)03 ceramics. 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

HCAPLUS COPYRIGHT 2003 ACS on STN 2003:50412 HCAPLUS L23 ANSWER 5 OF 44

ACCESSION NUMBER:

DOCUMENT NUMBER: 138:194073

Dependence of thermal conductivity of porous silicon TITLE: on porosity characterized by photoacoustic technique

AUTHOR(S): Shen, Qing; Toyoda, Taro

Department of Applied Physics and Chemistry, The CORPORATE SOURCE:

University of Electro-Communications, Tokyo, 182-8585,

Japan

SOURCE: Review of Scientific Instruments (2003), 74(1, Pt. 2),

601-603

CODEN: RSINAK; ISSN: 0034-6748 American Institute of Physics

PUBLISHER: DOCUMENT TYPE: Journal

LANGUAGE: English We have applied a photoacoustic (PA) technique to study the thermal properties of porous silicon (PS) films formed on p-type Si substrates by

electrochem. anodic etching with greatly different porosities (20-60%). From the dependence of the PA signals on the modulation frequency of excitation light measured under a

transmission detection configuration, the effective thermal diffusivities for the two-layer PS on Si samples were detd. Using a two-layer model, the thermal cond. of the PS films was evaluated and found to decrease greatly as the porosity is larger than 30%, which was two or three orders of magnitude lower than that reported for monocryst. silicon. The results demonstrate that the thermal cond. of PS films depends strongly on its

structure.

THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 15 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2002:825886 HCAPLUS

DOCUMENT NUMBER: 138:65644

Spectroelectrochemical sensing based on multimode TITLE: selectivity simultaneously achievable in a single

device. 12. Characterization of a channel waveguide

sensor

Ross, Susan E.; Seliskar, Carl J.; Heineman, William AUTHOR(S):

R.; Aryol, Saroj; Nevin, Joseph H.

Department of Chemistry, University of Cincinnati, CORPORATE SOURCE:

Cincinnati, OH, 45221, USA

Proceedings - Electrochemical Society (2001), SOURCE:

2001-18 (Chemical and Biological Sensors and Analytical

Methods II), 499-510

CODEN: PESODO; ISSN: 0161-6374

Electrochemical Society PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

Incorporation of planar channel waveguide technol. into a spectroelectrochem. sensor is described. In this sensor design, a potassium ion-exchanged BK7 glass channel waveguide was made by established photolithog. techniques. A pair of electrodes was placed along side the channel in a buss-bar configuration. A chem.-selective film was then spin-coated into the channel. Sensing was based on the change in attenuation of light propagated through the channel waveguide resulting from an optically absorbing analyte (ferricyanide-ferrocyanide couple) in the selective film. By applying either a triangular or square wave excitation potential waveform, electromodulation of the optical signal was demonstrated with Fe(CN)6-3/-4 as a model

electroactive couple that partitions into a PDMDAAC-SiO2 film {where PDMDAAC = poly(dimethyldiallylammonium chloride)} and absorbs at 442 nm. THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 15 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 7 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2002:788065 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 138:115567

Characterization of thermal properties of porous TITLE: silicon film/silicon using photoacoustic technique

Shen, Q.; Toyoda, T. AUTHOR(S):

Department of Applied Physics and Chemistry, The CORPORATE SOURCE:

University of Electro-Communications, Chofu, Tokyo,

182-8585, Japan

Journal of Thermal Analysis and Calorimetry (2002), SOURCE:

69(3), 1067-1073 CODEN: JTACF7; ISSN: 1418-2874

Kluwer Academic Publishers PUBLISHER:

DOCUMENT TYPE: Journal English LANGUAGE:

We have applied photoacoustic (PA) technique to study the thermal properties of porous silicon (PS) films formed on p-type Si substrates by electrochem. anodic etching. Four PS samples with close thicknesses but greatly different porosities (from 20 to 60%) were examd. From the dependences of the PA signals on the modulation

frequency of excitation light measured under a transmission detection configuration (TDC), effective thermal diffusivities for the two-layered PS/Si samples were detd. and found to decrease greatly from 0.095 to 0.020

cm2 s-1 as the porosity increased from 20 to 60%.

REFERENCE COUNT: THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS 11 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 8 OF 44 ACCESSION NUMBER: HCAPLUS COPYRIGHT 2003 ACS on STN

2002:641138 HCAPLUS

DOCUMENT NUMBER: 139:12411

A "virtual" electroanalytical instrument for square TITLE:

wave voltammetry

Economou, A.; Bolis, S. D.; Efstathiou, C. E.; AUTHOR(S):

Volikakis, G. J.

Department of Chemistry, Laboratory of Analytical CORPORATE SOURCE:

Chemistry, Aristotle University of Thessaloniki,

Thessaloniki, 540 06, Greece

Analytica Chimica Acta (2002), 467(1-2), 179-188 SOURCE:

CODEN: ACACAM; ISSN: 0003-2670

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

In this work we report the design, construction, control and applications of a computerized portable analyzer suitable for anal. square wave (SW) voltammetry. The analyzer is based on a home-made miniature potentiostat powered by two 9 V batteries. The electrochem. control program, developed in LabVIEW 5.1, generated the SW modulation signal, scanned the potential of the working electrode, acquired the current and displayed the results in the form of a voltammogram. Suitable programs were also developed in LabVIEW 5.1 for expts. involving preconcn. and for computer-assisted optimization of the instrumental parameters. The small dimensions, battery operation and inexpensive electronic components make this electrochem. analyzer ideal for in situ or field measurements. The utility of the instrument was assessed for different variants of SW voltammetry (direct SW voltammetry, anodic SW stripping voltammetry, cathodic SW stripping voltammetry, adsorptive SW stripping voltammetry, enrichment SW voltammetry) on different electrodes and applied to the detn. of riboflavin in multivitamin tablets.

THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 14 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER (9) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2000:801175 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 134:63182

Influence of electrolyte movement on measured TITLE:

electrochemical noise

AUTHOR(S): Legat, A.

Slovenian National Building and Civil Engineering CORPORATE SOURCE: '

Institute, Ljubljana, 1000, Slovenia

Corrosion (Houston) (2000), 56(11), 1086-1092 SOURCE:

CODEN: CORRAK; ISSN: 0010-9312

NACE International PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

The influence of movement of soln. on the characteristics of measured electrochem. noise (EN) was studied. For this reason,. measurements were performed in still soln., as well as in movement of the electrolyte. To relate measured EN to the development of corrosion processes, digitized images of electrodes were recorded continuously during these measurements. The characteristics of EN, in general, change significantly with strong movement of the soln.: stirring or laminar flow. Results of this study confirmed that the main source for this change is transformation of corrosion processes: in a still soln., the corrosion process tends to be localized, whereas during strong movement of the electrolyte, this tendency is oriented toward uniform corrosion. direct effect of the electrolyte movement on measured EN ( modulation of signals caused by spatiotemporal flow

disturbances) is small.

THERE ARE 32 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 32 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER (10) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2000:742444 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 133:358625

TITLE: Spectroelectrochemical Sensing Based on Multimode Selectivity Simultaneously Achievable in a Single

Device. 9. Incorporation of Planar Waveguide

Technology

Ross, Susan E.; Seliskar, Carl J.; Heineman, William AUTHOR(S):

Department of Chemistry, University of Cincinnati, CORPORATE SOURCE:

Cincinnati, OH, 45221-0172, USA Analytical Chemistry (2000), 72(22), 5549-5555 CODEN: ANCHAM; ISSN: 0003-2700 SOURCE:

American Chemical Society PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

Incorporation of planar waveguide technol. into a spectroelectrochem. sensor is described. In this sensor design, a potassium ion-exchanged BK7 glass waveguide was over-coated with a thin film of indium tin oxide (ITO) that served as an optically transparent electrode. A chem. selective film was spin-coated on top of the ITO film. The sensor supported five optical modes at 442 nm and three at 633 nm. Investigations on the impact of the ITO film on the optical properties of the waveguide and on the spectroelectrochem. performance of the sensor are reported. Sensing was based on the change in attenuation of light propagated through the waveguide resulting from an optically absorbing analyte. By applying either a triangular or square wave excitation potential waveform, electromodulation of the optical signal was demonstrated with Fe(CN)63-/4- as a model electroactive couple that partitions into a PDMDAAC-SiO2 film {where PDMDAAC =

poly(dimethyldiallylammonium chloride)} and absorbs at 442 nm. THERE ARE 51 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 51 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 11 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

2000:646852 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 133:314914

TITLE: Electrochemical and electrochemically modulated

reflectance AC voltammetry studies of electron

transfer kinetics between attached redox centers and a

mirror gold electrode

Brevnov, Dmitri A.; Finklea, Harry O. AUTHOR(S):

Department of Chemistry, West Virginia University, CORPORATE SOURCE:

Morgantown, WV, 26506, USA

Journal of the Electrochemical Society (2000), 147(9), SOURCE:

3461-3466

CODEN: JESOAN; ISSN: 0013-4651

Electrochemical Society PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

Small amplitude electrochem. a.c. voltammetry (ACV) is employed to study kinetics of electron transfer between Ru redox centers attached to the electrode surface via C10 alkanethiols and gold mirror electrodes. The equations for faradaic admittance of strongly adsorbed electroactive species in the case of a Langmuir isotherm are applied to det. the total coverage of redox centers (.theta.total) and kinetic parameters: the std. rate const. (ks) and transfer coeff. The ks appears to increase and .theta.total appears to decrease as the perturbation frequency increases: in a sep. expt., large amplitude ACV was performed simultaneously with electrochem. modulated reflectance a.c. voltammetry (EMR ACV) on the same electroactive monolayer/electrode system. The electromodulation reflectance coeff. (X) is defined in the frequency domain as a ratio of a.c. electroreflectance to both d.c. electroreflectance and the interfacial a.c. potential (Einter). The X is a more useful quantity than the ratio of a.c. electroreflectance to d.c. electroreflectance for representation of the electroreflectance data. The X is exactly out-of-phase with the faradaic admittance at the wavelength region (410-440 nm) corresponding to the absorption band of the reduced form of the Ru complex. Therefore, the a.c. electroreflectance **signal** is dominated by the **modulation** of the electrode coverage of a given redox state with Einter. Electrochromic effects are negligible. Accurate calcn. of ks from EMR ACV data is complicated because of the nonlinear relation between the faradaic a.c. current and large amplitude Einter. Thus, under the authors' exptl. conditions, ACV is the preferred method to det. ks.

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER (12 of 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1999:805690 HCAPLUS

DOCUMENT NUMBER: 132:95725

Influence of palladium films on hydrogen gas entry TITLE:

into iron: a study by electrochemical impedance

spectroscopy

Bruzzoni, P.; Carranza, R. M.; Lacoste, J. R. Collet AUTHOR(S):

Comision Nacional de Energia Atomica, Buenos Aires, CORPORATE SOURCE:

1429, Argent.

SOURCE: International Journal of Hydrogen Energy (1999),

> Volume Date 2000, 25(1), 61-65 CODEN: IJHEDX; ISSN: 0360-3199

Elsevier Science Ltd.

PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English



Covering the bare surface of a hydride forming metallic material with a thin film of Pd is a method to accelerate hydriding, through avoiding the formation of a passive oxide film and allowing hydrogen to diffuse easily through the Pd film. We performed permeation expts. to study the hydrogen ingress rate into Pd-plated iron membranes. The exptl. conditions are such that the Pd deposits and their interfaces (rather than the bulk iron substrate) det. the permeation results. Thus, it is possible to characterize the Pd film behavior with respect to the passage of hydrogen. We used the electrochem. impedance spectroscopy technique, where the excitation signal was a modulated hydrogen pressure, and the response signal was the modulated hydrogen flux, measured by an electrochem. method at the exit side of the permeation membrane.

THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER (13 )OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1999:559312 HCAPLUS

DOCUMENT NUMBER: 131:266213

Development of new spectroelectrochemical waveguide TITLE:

sensors

Ross, Susan E.; Seliskar, Carl J.; Heineman, William AUTHOR(S):

Dept. of Chemistry, University of Cincinnati, CORPORATE SOURCE:

Cincinnati, OH, 45221-0172, USA

Polymer Preprints (American Chemical Society, Division SOURCE:

of Polymer Chemistry) (1999), 40(2), 1265-1266

CODEN: ACPPAY; ISSN: 0032-3934

American Chemical Society, Division of Polymer PUBLISHER:

Chemistry

DOCUMENT TYPE: Journal English LANGUAGE:

The authors describe the combination of a new sensor concept employing 3

modes of selectivity and a novel spectroelectrochem. sensor design based on planar waveguide technol. The modulated optical

signal correlated to the electrochem. in the

chem.-selective film and that the film was crucial in preconcg. sufficient amt. of analyte near the waveguide surface for detection. Optical detection employing the simultaneous trimodal selectivity concept was achieved with an integrated optical waveguide. By optimizing the film deposition and optical coupling, sensitivity and detection limits should

be significantly improved.

THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 11 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 14 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1999:539684 HCAPLUS ACCESSION NUMBER:

Strategies for achieving selectivity in chemical TITLE:

sensors.

Heineman, William R.; Halsall, H. Brian; Seliskar, AUTHOR(S):

Carl J.

Department of Chemistry, University of Cincinnati, CORPORATE SOURCE:

Cincinnati, OH, 45221-0172, USA

Book of Abstracts, 218th ACS National Meeting, New SOURCE:

Orleans, Aug. 22-26 (1999), ANYL-120. American

Chemical Society: Washington, D. C.

CODEN: 67ZJA5

Conference; Meeting Abstract DOCUMENT TYPE:

LANGUAGE: English

Examples of strategies we have used for the development of new chem. sensors will be discussed. The emphasis will be on achieving adequate selectivity for sensing in complex samples. One example is electrochem. immunoassay based on labels that are electroactive or

catalyze the prodn. of an electroactive product. A microcapillary immunoreactor with electrochem. detection has led to the achievement of detection limits as low as the zeptomole range. This concept is being extended to the development of an electrochem. immunosensor on a single microchip. A sensor for in vivo monitoring of a heart imaging agent for nuclear medicine is a second example that illustrates how detection limit and selectivity can be improved with polymer-modified electrodes. A sensor has been developed for in vivo monitoring the prototype heart imaging agent Re(DMPE), where DMPE.ident.1,2-bis(dimethyl-phosphino)ethane. A third example is a sensor that provides three levels of selectivity by the combination of selective partitioning into a film with monitoring of an optical signal that is electrochem. modulated.

L23 ANSWER (15) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1999:99774 HCAPLUS

DOCUMENT NUMBER: 130:262507

TITLE: Age-related changes in the capacity, rate, and

modulation of dopamine uptake within the striatum and

nucleus accumbens of Fischer 344 rats: an in vivo

electrochemical study

AUTHOR(S): Hebert, Meleik A.; Gerhardt, Greg A.

CORPORATE SOURCE: Neuroscience Training Program, Rocky Mountain Center

for Sensor Technology, University of Colorado Health

Sciences Center, Denver, CO, USA

SOURCE: Journal of Pharmacology and Experimental Therapeutics

(1999), 288(2), 879-887

CODEN: JPETAB; ISSN: 0022-3565

PUBLISHER: American Society for Pharmacology and Experimental

Therapeutics

DOCUMENT TYPE: Journal

LANGUAGE: English Age-related changes in the capacity, rate, and modulation of dopamine (DA) uptake within the striatum and the nucleus accumbens core of Fischer 344 rats were investigated using in vivo electrochem. recordings coupled with local drug application techniques. Equimolar amts. of DA were pressure ejected into the striatum and the nucleus accumbens of 6-, 12-, 18-, and 24-mo old rats. The DA ejections produced larger DA signal amplitudes in the older rats, suggesting age-related differences in the capacity to clear extracellular DA. Within the striatum, the capacity and rate of DA uptake were reduced by 50% in the aged groups (18 and 24 mo) compared with the younger rats (6 and 12 mo). In the nucleus accumbens, significant redns. in DA uptake capacity and rate were obsd. in the 24-mo group. In both brain regions and in all age groups studied, the rate of DA uptake was found to be concn.-dependent until a maximal rate was reached. The max. rate of DA transport was significantly reduced in both the striatum and the nucleus accumbens of aged rats (18 and 24 mo vs. 6 and 12 mo). The ability of nomifensine, an inhibitor of the DA transporter, to modulate DA signal amplitudes in the striatum and the nucleus accumbens was also decreased with age (24 mo vs. 6 mo). Taken together, these findings demonstrate substantial age-related deficits in DA uptake processes within the striatum and the nucleus accumbens, consistent with the hypothesis that DA uptake may be slowed in

aged animals to compensate for redns. in DA release.

REFERENCE COUNT: 41 THERE ARE 41 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 16 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1999:17731 HCAPLUS

DOCUMENT NUMBER: 130:176753

TITLE: Tailoring Perfluorosulfonated Ionomer-Entrapped Sol-Gel-Derived Silica Nanocomposite for Spectroelectrochemical Sensing of Re(DMPE)3+

AUTHOR(S): Hu, Zhongmin; Slaterbeck, Andrew F.; Seliskar, Carl

J.; Ridgway, Thomas H.; Heineman, William R.

CORPORATE SOURCE: Department of Chemistry, University of Cincinnati,

Cincinnati, OH, 45221-0172, USA Langmuir (1999), 15(3), 767-773 CODEN: LANGD5; ISSN: 0743-7463

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

SOURCE:

The influence of the initial molar ratio of H2O relative to tetra-Et orthosilicate (TEOS) precursor and the content of Nafion ionomer in sol-gel-derived SiO2 composites on the voltammetric response of electrodes modified with these composites for [ReI(DMPE)3]+ was studied. The slow diffusion of [ReI(DMPE)3]+ in Nafion can be significantly improved by dispersing Nafion in sol-gel-derived SiO2, and the diffusion of [ReI(DMPE)3]+ in such a composite increases with the increase in H2O/TEOS molar ratio and the decrease in Nafion content. With the mass ratio of Nafion relative to sol-gel-derived SiO2 being 40:100 and the initial molar ratio of H2O relative to TEOS being 20:1, the electrodes modified with the derived Nafion-SiO2 nanocomposite exhibited an apparent peak current increase rate, during preconcn. of [ReI(DMPE)3]+, that was approx. three times faster than the corresponding Nafion-modified electrode. Compared with bare In-Sn oxide (ITO) glass, the composite-coated ITO glass showed a 25-fold enhancement in voltammetric response to [ReI(DMPE)3]+. The suitability of the developed optically transparent Nafion-SiO2 composite for **spectroelectrochem**. sensing of [ReI(DMPE)3]+ was demonstrated. The [ReI(DMPE)3]+ extd. into the coating (.apprx.0.4 .mu.m in thickness) was electrolyzed to [ReII(DMPE)3]2+. Under attenuated total reflection mode, the in-situ electrogenerated chromophore [ReII(DMFE)3]2+ was monitored by probing its interaction with the evanescent field of light of a selected wavelength. Thus, the elements required for a spectroelectrochem. sensor with three modes of selectivity were demonstrated: partitioning into the film on an electrode surface and an

electrochem. modulated optical signal.

REFERENCE COUNT: 43 THERE ARE 43 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER (17) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1998:542292 HCAPLUS

DOCUMENT NUMBER: 129:226042

TITLE: Electrical stimulation of the prefrontal cortex increases cholecystokinin, glutamate, and dopamine

release in the nucleus accumbens: an in vivo microdialysis study in freely moving rats

AUTHOR(S): You, Zhi-Bing; Tzschentke, Thomas M.; Brodin, Ernst;

Wise, Roy A.

CORPORATE SOURCE: Center for Studies in Behavioral Neurobiology,

Department of Psychology, Concordia University,

Montreal, QC, H3G 1M8, Can.

SOURCE: Journal of Neuroscience (1998), 18(16), 6492-6500

CODEN: JNRSDS; ISSN: 0270-6474

PUBLISHER: Society for Neuroscience

DOCUMENT TYPE: Journal LANGUAGE: English

In vivo microdialysis, RIA, and HPLC with **electrochem**. or fluorometric detection were used to investigate the release of cholecystokinin (CCK), glutamate (Glu), and dopamine (DA) in nucleus accumbens septi (NAS) as a function of ipsilateral elec. stimulation of medial prefrontal cortex (mPFC). CCK was progressively elevated by mPFC stimulation at 50-200 Hz. Stimulation-induced CCK release was intensity-dependent at 250-700 .mu.A. NAS Glu and DA levels were each elevated by stimulation at 25-400 Hz; the dopamine metabolites DOPAC and homovanillic acid were increased by stimulation at 100-400 Hz. When rats

were trained to lever press for mPFC stimulation, the stimulation induced similar elevations of each of the three transmitters to those seen with experimenter-administered stimulation. Perfusion of 1 mM kynurenic acid (Kyn) into either the ventral tegmental area (VTA) or NAS blocked lever pressing for mPFC stimulation. VTA, but not NAS, perfusion of Kyn significantly attenuated the increases in NAS DA levels induced by mPFC stimulation. Kyn did not affect NAS CCK or Glu levels when perfused into either the VTA or NAS. The present results are consistent with histochem. evidence and provide the first in vivo evidence for the existence of a releasable pool of CCK in the NAS originating from the mPFC. Although dopamine is the transmitter most closely linked to reward function, it was CCK that showed frequency-dependent differences in release corresponding most closely to rewarding efficacy of the stimulation. Although not essential for the reward signal itself, coreleased CCK may

modulate the impact of the glutamatergic action in this behavior.
REFERENCE COUNT: 98 THERE ARE 98 CITED REFERENCES AVAILABLE FOR THIS
RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 18 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1998:180215 HCAPLUS

DOCUMENT NUMBER: 128:327865

TITLE: Anodic electrodeposition of iridium oxide films

AUTHOR(S): Petit, Michel A.; Plichon, Vincent

CORPORATE SOURCE: URA 429 C.N.R.S., Laboratoire de Chimie et Electrochimie des Materiaux Moleculaires, Ecole

Superieure de Physique et Chimie Industrielles, Paris,

75231, Fr.

SOURCE: Journal of Electroanalytical Chemistry (1998), 444(2),

247-252

CODEN: JECHES; ISSN: 0368-1874

Elsevier Science S.A.

DOCUMENT TYPE: Journal LANGUAGE: English

PUBLISHER:

IrOx films were deposited on conducting transparent SnO2-coated glass substrates with sheet resistance of .apprx.10 .OMEGA./.box. by electrochem. oxidn. of sol. iridium species in a basic aq. medium. Before being efficient for deposition, the iridium-contg. solns. have to be aged for a min. time which is detd. by UV-visible spectrophotometry. The IrOx films prepd. following the procedure described show cyclic voltammograms similar to those of anodic iridium oxide films (AIROF)s grown on iridium surfaces, i.e. one or two redox systems depending on whether the pH of the electrolyte is lower or >5. The IrOx films could be completely discolved in beiling hydrochloric acid solns, and consequently

grown on iridium surfaces, i.e. one or two redox systems depending on whether the pH of the electrolyte is lower or >5. The IrOx films could be completely dissolved in boiling hydrochloric acid solns. and consequently their iridium contents were very precisely detd. These anal. clearly showed that a total of two electrons/iridium are exchanged in a basic medium, i.e. one electron/iridium for each redox system as the charges involved in both systems are nearly the same. The optical properties of the IrOx films were monitored in situ under potential cycling. The more cathodic redox system was assigned to Ir(III)/Ir(IV), it exhibits an electrochromic behavior characterized by a complete transparency of the reduced state and a coloration factor approaching 18 cm2 C-1 at 633 nm.

However, no significant modulation of the transmitted signal at 633 nm was detected for the 2nd redox system which was only obsd. in basic media and attributed to Ir(IV)/Ir(V).

REFERENCE COUNT: 25 THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 19 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1997:777782 HCAPLUS

DOCUMENT NUMBER: 128:94489

TITLE: In situ infrared spectroelectrochemistry: applications

to electrocatalysis

AUTHOR(S): Lopes, M. I. S.; Proenca, L.

CORPORATE SOURCE: CECUL, Faculdade de Ciencias da Universidade de

Lisboa, Lisbon, 1294, Port.

Portugaliae Electrochimica Acta (1997), 15(June), SOURCE:

81-111

CODEN: PEACEZ; ISSN: 0872-1904 Portugaliae Electrochimica Acta

DOCUMENT TYPE: Journal; General Review

English LANGUAGE:

PUBLISHER:

The use of in situ IR vibrational spectroscopy on the elucidation at a mol. level of electrochem. induced phenomena, i.e., in situ IR spectroelectrochem., was rapidly developing in the last two decades, since a pioneer work of Bewick et al., by using modulation and synchronous detection of the signal in external reflectance IR spectroscopy. In situ IR spectroelectrochem. is a powerful tool for study of the electrode/soln. interface, by providing information on the structural, bonding and dynamical properties of adsorbate species on electrode surfaces, on the behavior of supporting electrolyte and solvent mols. at the electrochem. interface and also on reaction mechanisms and kinetics of electrocatalytic reactions. An overview of some theor. aspects of IR reflectance spectroscopy is presented and exptl. details of some in situ techniques are described, EMIRS (Electrochem. Modulated IR Reflectance Spectroscopy), SNIFTIRS (Subtractively Normalized Interfacial FTIR Reflectance Spectroscopy) and SPAIRs (Single Potential Alteration IR Reflectance Spectroscopy). A few selected application of these techniques to electrocatalysis are also presented. A review with 31 refs.

THERE ARE 31 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 31 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER (20) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1997:663877 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 128:9816

Hydrodynamic modulation using vibrating electrodes: TITLE:

Application to electroanalysis

Williams, David E.; Ellis, Keith; Colville, Alistair; AUTHOR(S):

Dennison, Stephen J.; Laguillo, George; Larsen, Jane

Department of Chemistry, University College London, 20 Gordon St, London, WC1H OAJ, UK CORPORATE SOURCE:

Journal of Electroanalytical Chemistry (1997), SOURCE:

432(1-2), 159-169

CODEN: JECHES; ISSN: 0368-1874

Elsevier PUBLISHER: DOCUMENT TYPE: Journal LANGUAGE: English

Microband (10 .mu.m.times.2 mm), multiple parallel microband (spacing 20 .mu.m) and planar (.apprx.1 mm2) electrodes, fabricated by screen printing and vibrated (2-50 Hz; 1-2 mm amplitude) either continuously or in pulsed motion parallel to the short axis of the band and perpendicular to the face of the planar electrodes, are described. The band electrodes were made by printing successive layers of gold and insulator onto a ceramic substrate then snapping along a pre-scribed line to expose a fresh surface. A clean and repeatable electrode surface for a disposable device was obtained by this simple expedient. The resulting microband electrodes when vibrated showed the signal enhancement characteristic of vibrated wire electrodes, the modulation depth and signal enhancement being smaller than those characteristic of a vibrated microwire but larger than those from larger diam. wires. The printed bands have the major advantages of ease of prepn. and use and of low cost and repeatability in bulk manuf. Reagents were also printed onto the devices and were rapidly dissolved and mixed into the test soln. by the vibration. A low-cost, single-use device for anal., without prior calibration, of chlorine in water was demonstrated with detection limit

around 0.3 .mu.mol dm-3 (0.02 ppm) and linearity into the mmol dm-3 range. Multiple microband electrodes were used to implement, in a simple and inexpensive way, generator-collector methods of electrochem. titrn., exemplified by the detn. of ascorbic acid using electrogenerated Fe(III). The anal. methods are repeatable and practical, although the hydrodynamics of these systems are complex. The current to planar electrodes oriented vertically and vibrated along the surface normal is not modulated: a uniform circulating flow appears to be set up. The current to microband electrodes vibrated parallel to the short axis is strongly modulated. For sinusoidal motion the current waveform is an amplitude- and frequency-dependent superposition of a no. of waves, each perfectly periodic in a simple multiple of the vibration frequency. The mean and max. current are independent of the amplitude of the motion. qual. description of the hydrodynamics of the vibrating electrodes is given and the predicted dependence of mean current on (vibration frequency) 1/2 demonstrated, despite the complexity of the current waveforms. The idea of using pulsed motion to renew periodically the concn. boundary conditions is discussed briefly.

THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 35 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L23 ANSWER 21 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1997:358237 HCAPLUS

127:127811 DOCUMENT NUMBER:

TITLE: Two-dimensional infrared correlation analysis of

electrochemical reactions

Osawa, Masatoshi; Yoshii, Katsumasa; Hibino, Yu-ichi; AUTHOR(S):

Nakano, Tatsuhiko; Noda, Isao

Catalysis Research Center, Hokkaido University, CORPORATE SOURCE:

Sapporo, Japan

Journal of Electroanalytical Chemistry (1997), SOURCE:

426(1-2), 11-16

CODEN: JECHES; ISSN: 0368-1874

Elsevier PUBLISHER: DOCUMENT TYPE: Journal English LANGUAGE:

Three advanced IR spectroscopy techniques, surface-enhanced IR absorption spectroscopy (SEIRAS), step-scan Fourier-transform interferometry, and two-dimensional (2D) IR correlation anal., were applied to the study of **electrochem**. reactions. A combined use of SEIRAS and step-scan interferometry enables time-resolved spectral monitoring of electrochem. reactions with time-resolns. ranging from microseconds to milliseconds. Two-dimensional-IR correlation anal. highlights the dynamic information obscured in the time-resolved spectra. The basic concept of 2-dimensional-IR is somewhat analogous to that of 2-dimensional-NMR, and synchronous and asynchronous 2-dimensional spectra defined by two independent wavenumbers are generated by a correlation anal. of dynamic fluctuation of IR signals induced by a potential modulation. The synchronous and asynchronous spectra characterize the coherence and incoherence resp. of dynamic fluctuations of IR signals at two different wavenumbers. Bands arising from different transient species are clearly differentiated by their characteristic time-dependent behavior. The temporal relation between the intensity fluctuations of different bands also becomes clear. Deep insights into reaction processes are gained. The utilities of 2-dimensional-IR are demonstrated for the 1-electron redn. of heptylviologen at a silver electrode surface.

L23 ANSWER(22)OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1996:32775 **HCAPLUS** ACCESSION NOMBER:

DOCUMENT NUMBER: 124:156822

Comparison between Real-Time Polarization Modulation TITLE: and Static Linear Polarization for in Situ Infrared

Spectroscopy at Electrode Surfaces

Richmond, William N.; Faguy, Peter W.; Jackson, AUTHOR(S):

Richard S.; Weibel, Stephen C.

Department of Chemistry, University of Louisville, CORPORATE SOURCE:

Louisville, KY, 40292, USA

Analytical Chemistry (1996), 68(4), 621-8 CODEN: ANCHAM; ISSN: 0003-2700 SOURCE:

American Chemical Society PUBLISHER:

DOCUMENT TYPE: Journal English LANGUAGE:

Both conventional static linear polarization selection and a new real-time sampling technique for polarization modulation were used for in situ FTIR spectroelectrochem. studies of copper electrodes. Three aq. probe species were studied: thiocyanate ion, imidazole, and glucose. data were obtained under identical electrochem. conditions for the systems studied, a direct comparison of the optical methodologies was possible. The modulation technique provided excellent rejection of signals due to atm. gases and increases in overall sensitivity relative to the static method. Both techniques provided equiv. information with respect to surface species, but differed in the extent to which the electrolyte soln. was sampled. Thus, by using both techniques to study a given electrochem. process, a means to discriminate spectral features due to adsorbed species from those due to soln.-phase components is established.

L23 ANSWER (23) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1995:802881 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 123:240123

TITLE: Charge transfer at photoelectrochemical solar cells: a

comparison of data from impedance and light-modulated

spectroscopy

Schefold, J. AUTHOR(S):

Institut fuer Physikalische Elektronik, Universitaet CORPORATE SOURCE:

Stuttgart, Stuttgart, 70569, Germany

Journal of Electroanalytical Chemistry (1995), SOURCE:

394(1-2), 35-48 CODEN: JECHES; ISSN: 0368-1874

PUBLISHER: Elsevier DOCUMENT TYPE: Journal LANGUAGE: English

The impedance and the response to a light-modulated signal are compared at p-InP|V3+/2+ semiconductor|liq. junctions. The familiar recombination semicircle, seen in the complex-plane representation of the light-modulated response, is explained by electrochem. charge transfer (CT) phenomena. Transformable differential models are proposed which fit impedance and light-modulated data at bare and Ag-coated p-InP. Both spectroscopic methods verify an increase in the CT rate (the reason for higher solar-cell efficiency) and the capacity of the interface owing to the Ag coating. Unlike those at Aq-coated p-InP, CT rates measured at the open-circuit potential at bare electrodes show a dependency on illumination intensity. Though both spectroscopic methods are based on identical theor. principles, the no. of parameters which can be detd. at the system discussed is different (4 parameters in impedance, 2 in the light-modulated method). Capacity data reported from several anal. methods are summarized and attributed to the same phys. quantity.

L23 ANSWER 24 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1995:362621 HCAPLUS

DOCUMENT NUMBER: 122:304510

Tip-position modulation and lock-in detection in TITLE:

scanning electrochemical microscopy

INVENTOR(S): Bard, Allen J.; Wipf, David O.

University of Texas System, USA PATÊNT ASSIGNEE(S):

SOURCE: U.S., 22 pp. CODEN: USXXAM

DOCUMENT TYPE: Patent English LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_ \_\_\_\_\_ \_\_\_\_\_ US 5382336 US 1992-869301 19920416 Α 19950117 PRIORITY APPLN. INFO.: US 1992-869301 19920416

Small-amplitude tip-position modulation (TPM) in combination with lock-in detection of the modulated current signal is applied to a scanning electrochem. microscope (SECM) to improve signal-to-noise ratio and to enhance image resoln. Phase shift information from the a.c. TPM signal and the bipolar response of TPM over insulators and conductors make TPM-SECM superior to conventional SECM imaging methods as well as useful for measuring substrate cond.

L23 ANSWER (25) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1994:469103 HCAPLUS ACCESSION NUMBER:

121:69103 DOCUMENT NUMBER:

Multiple source and detection frequencies in detecting TITLE: atomic or molecular spectra and/or threshold phenomena

associated with the same

Weiss, Paul S.; Stranick, Stephan J. INVENTOR(S):

Penn State Research Foundation, USA; Biotechnology PATENT ASSIGNEE(S):

Research and Development Corp.

SOURCE: PCT Int. Appl., 48 pp.

CODEN: PIXXD2

Patent DOCUMENT TYPE: English LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_\_ \_\_\_\_\_ \_\_\_\_\_ ----WO 1993-US6720 19930716 WO 9402841 A1 19940203 WO 1993-US6720 19930716 W: CA, JP, KR RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE US 1992-916165 19920717 A 19931207 US 5268573 19940125 US 1992-979597 19921120 Α US 5281814 US 5397896 US 1993-92016 19930715 Α 19950314 US 5434842 US 1993-92133 19930715 19950718 Α US 1992-916165 A 19920717 PRIORITY APPLN. INFO.: US 1992-979597 A 19921120 US 1993-56348 A 19930430 US 1993-92016 A 19930715 US 1993-92133 A 19930715

Multiple frequency sources of light are used to apply a light signal to an AB app. (e.g., a scanning near-field optical microscope or a STM). After being modulated by the sample, the signal is detected and a spectrometer which detects simultaneously at the multiple input frequencies and combinations thereof. This permits multiple substances to be monitored simultaneously. When applied to electrochem. cells, this invention is particularly useful for measuring dynamic information of the electrochem. cell. When applied to a Coulomb blockade device, charging threshold and charge dissipation rates can be

L23 ANSWER (26 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN 1993:681758 HCAPLUS ACCESSION NUMBER:

119:281758 DOCUMENT NUMBER:

Step-scanning interferometer with digital signal TITLE:

processing

Manning, Christopher J.; Griffiths, Peter R. AUTHOR(S):

Dep. Chem., Univ. Idaho, Moscow, ID, 83844-2343, USA CORPORATE SOURCE:

Applied Spectroscopy (1993), 47(9), 1345-9 CODEN: APSPA4; ISSN: 0003-7028 SOURCE:

DOCUMENT TYPE: Journal LANGUAGE: English

A novel step-scan FT-IR spectrometer incorporating a digital signal processor for demodulation of the detector

signal is described. The potential advantages of this method of signal processing are discussed and illustrated. The instrument is based on a com. cube-corner interferometer which has been modified by replacement of the drive motor with a stepper motor-micrometer and piezoelec. transducer combination. The interferometer retardation is feedback controlled by a 486-50 personal computer, which also controls the digital signal processor and collects spectral data. More than one phase modulation frequency can be imposed simultaneously, allowing for a multiplex advantage in photoacoustic depth profiling. Digital signal processing allows for simultaneous demodulation of multiple frequencies which would normally require several lock-in

amplifiers. Data that illustrate the feasibility of these concepts are presented. The suitability of this instrument for double-modulation step-scan FT-IR measurements such as polymer stretching and electrochem. modulated step-scan FT-IR is also discussed.

L23 ANSWER (27) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1993:594622 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 119:194622

Step-scan FT-IR external reflection spectrometry of TITLE:

the electrode/electrolyte interface by potential

modulation

Budevska, Boiana O.; Griffiths, Peter R. AUTHOR(S):

Dep. Chem., Univ. Idaho, Moscow, ID, 83843, USA CORPORATE SOURCE: Analytical Chemistry (1993), 65(21), 2963-71 CODEN: ANCHAM; ISSN: 0003-2700 SOURCE:

DOCUMENT TYPE: Journal English LANGUAGE:

The applications of conventional rapid-scanning Fourier transform IR (FT-IR) spectrometry for IR spectroelectrochem. measurements involving potential modulation is limited by the cross talk between the Fourier frequencies and the frequency at which the potential is modulated. In this report the authors show the first demonstration of

electrochem. modulated IR spectrometry using a step-scanning phase-modulated interferometer. The wavenumber of the absorption band of

CO adsorbed on polycryst. Pt electrode was modulated through the electrochem. Stark effect, and the detector signal was

demodulated using two lock-in amplifiers, one for the phase modulation and the other for the potential modulation. The optimal data collection parameters and data manipulation techniques are discussed. The dynamic information contained in the in-phase and quadrature spectra is found to provide the time lag of the abs. potential at the surface of the working electrode relative to the applied potential. This parameter is an important characteristic of an electrochem. cell that is not

readily accessible by other means.

L23 ANSWER 28 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN 1993:134711 HCAPLUS ACCESSION NOMBER:

DOCUMENT NUMBER: 118:134711

Experimental implementation and capabilities of TITLE:

Fourier transform hydrodynamic modulation voltammetry

Schwartz, Daniel T. AUTHOR(S):

Dep. Chem. Eng., Univ. Washington, Seattle, WA, 98195, CORPORATE SOURCE:

Journal of the Electrochemical Society (1993), 140(2), SOURCE:

452 - 8

CODEN: JESOAN; ISSN: 0013-4651

Journal DOCUMENT TYPE: LANGUAGE: English

The 1st exptl. implementation of Fourier transform hydrodynamic modulation voltammetry (FTHMV) is reported. Two different types of expts. were performed to illustrate the capabilities of the technique. First, the anal. utility of FTHMV is demonstrated by applying the technique to study the frequency response characteristics of mass-transfer limited redn. of 2 mM ferric ion at a speed-modulated rotating disk electrode. FTHMV, performed using a square wave rotation rate modulation, is a fast and accurate method for characterizing the system's frequency response in a single expt., and thus provides an excellent alternative to the common practice of performing many expts. at 1 frequency at a time. The 2nd application of FTHMV reported here is the anal. of nonlinearities in the mass-transfer limited current to a rotating disk electrode modulated at a single frequency. The nonlinear signal is proportional to the amplitude of the rotation rate modulation raised to the power of the harmonic component. Also discussed are the implications of time domain signal processing for improving the signal-to-noise performance of FTHMV.

L23 ANSWER 29 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1992:418523 HCAPLUS

DOCUMENT NUMBER: 117:18523

Scanning electrochemical microscopy. 15. TITLE:

Improvements in imaging via tip-position modulation

and lock-in detection

Wipf, David O.; Bard, Allen J. AUTHOR(S):

CORPORATE SOURCE: Dep. Chem. Biochem., Univ. Texas, Austin, TX, 78712,

USA

Analytical Chemistry (1992), 64(13), 1362-7 CODEN: ANCHAM; ISSN: 0003-2700 SOURCE:

Journal DOCUMENT TYPE: LANGUAGE: English

The use of small amplitude tip-position modulation (TPM) in combination with lock-in detection of the modulated current signal

greatly improves the sensitivity and image resoln. of the scanning electrochem. microscope and provides a method of distinguishing between conductive and insulating areas on the substrate surface being examd. The exptl. in-phase current vs. distance behavior is characterized for insulating and conducting surfaces for various modulation amplitudes and frequencies. A simple deriv. model of the d.c. response is adequate to derive the in-phase TPM response at conductors; the insulator response does not conform to the theory and is, in fact, more sensitive than predicted. Demonstration images of an interdigitated array electrode using the dc and in-phase TPM signal are compared. The in-phase TPM image is found to be superior for imaging because of its improved sensitivity to insulating surfaces and its bipolar response over insulators and conductors.

L23 ANSWER (30) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1991:459410 HCAPLUS

115:59410 DOCUMENT NUMBER:

TITLE: In-situ spectroelectrochemistry of adsorbed methylene

blue on a sulfur-modified gold electrode

Lezna, R. O.; De Tacconi, N. R.; Hahn, F.; Arvia, A. AUTHOR(S):

Fac. Cienc. Exactas, Univ. Nac. La Plata, La Plata,

CORPORATE SOURCE: 1900, Argent.

Journal of Electroanalytical Chemistry and Interfacial SOURCE:

Electrochemistry (1991), 306(1-2), 259-69

CODEN: JEIEBC; ISSN: 0022-0728

DOCUMENT TYPE: Journal English LANGUAGE:

The properties of a methylene blue layer adsorbed onto a sulfur-modified gold electrode have been studied by electrochem. and in-situ reflectance spectroscopy in the visible region. At satn. coverage, methylene blue monomer and dimer bands were obsd. in the double-layer region through the electrochromism of the adsorbed film arising from the effect of the double-layer field on the mol. transition moment. Possible dye orientations in the adsorbed state are discussed with the aid of the optical measurements. At low surface concns., the optical signal stems from modulation of the monomer/dimer interconversion. Repeated electrochem. redn. and oxidn. of adsorbed methylene blue result in gradual desorption of the dimer species, probably due to a change in the mol. geometry upon redn. which may evolve from a planar configuration to a folded structure.

L23 ANSWER 31 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1991:177401 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 114:177401

TITLE: Use of time-dependent chemical sensor signals for

selective identification

AUTHOR(S): Maclay, G. J.; Stetter, J. R.; Christesen, S. D. CORPORATE SOURCE: Chem. Res. Dev. Eng. Cent., Aberdeen Proving Ground,

MD, USA

SOURCE: Report (1989), CRDEC-TR-083; Order No. AD-A212729, 27

pp. Avail.: NTIS

From: Gov. Rep. Announce. Index (U. S.) 1990, 90(2),

Abstr. No. 002,706

DOCUMENT TYPE: Report LANGUAGE: English

A sensor is described for identifying and quantifying different components in a gas mixt. by decoding information from a modulated output signal. The system operates by passing the unknown gas into a reaction chamber contg. a heated catalytic filament. The output products of the chamber are detected by an electrochem. sensor. The concn. of the reaction products is modulated by varying the temp. of the catalytic filament. A theory is presented for analyzing the modulated output signal. The validity of the anal. is confirmed by expts. conducted with benzene, carbon monoxide, hydrogen cyanide, and by computer simulations. It appears possible to identify different species in the inlet gas stream if the species undergo chem. reactions in the reactor that have different activation energies. signal magnitude at peaks or valleys in the output is directly proportional to concns. even though a steady-state condition is not The approach may be generalized to other detector systems. reached.

L23 ANSWER (32) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1991:20622 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 114:20622

Homogeneous electrochemiluminescent specific-binding TITLE:

partner assay method, reagents, system, and kit, and

use of the method in immunoassays

Shah, Haresh P.; Hall, Lee O.; Powell, Michael J.; Massey, Richard J. INVENTOR(S):

IGEN Inc., USA PATENT ASSIGNEE(S):

SOURCE: PCT Int. Appl., 60 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 11

PATENT INFORMATION:

```
KIND DATE
                                        APPLICATION NO. DATE
    PATENT NO.
                    ----
     _____
                                        _____
                    A1 19900517
                                        WO 1989-US4919
                                                        19891031
    WO 9005301
        W: AU, DK, FI, JP, KR, NO
        RW: AT, BE, CH, DE, FR, GB, IT, LU, NL, SE
                   A1 19900528
B2 19930610
    AU 8946357
                                        AU 1989-46357
                                                        19891031
    AU 637775
    ZA 8908290
                                        ZA 1989-8290
                                                        19891031
                         19910626
                     Α
                                        EP 1989-913023
                                                        19891031
    EP 446245
                     A1
                        19910918
                         19990512
    EP 446245
                    В1
       R: AT, BE, CH, DE, FR, GB, IT, LI, LU, NL, SE
    JP 04502209
                     T2
                          19920416
                                        JP 1990-500352
                                                        19891031
    JP 2992974
                     В2
                          19991220
    IL 92164
                         19960119
                                        IL 1989-92164
                                                        19891031
                     A 1
                    A2
    EP 757252
                         19970205
                                        EP 1996-202510
                                                        19891031
    EP 757252
                    А3
                         19970416
        R: AT, BE, CH, DE, FR, GB, IT, LI, LU, NL, SE
    AT 180057
               E
                          19990515
                                        AT 1989-913023
                                                        19891031
                         19900503
                                        CA 1989-2002101 19891102
    CA 2002101
                     AA
    CA 2002101
                    С
                          20020528
    US 5635347
                    Α
                         19970603
                                        US 1994-188943
                                                        19940128
                                        US 1994-255824 19940608
    US 5705402
                    Α
                         19980106
                    B1
                          20020910
                                        US 1994-335183
                                                        19941107
    US 6448091
                                        US 1994-348749
    US 5770459
                     Α
                          19980623
                                                        19941201
    US 5779976
                                        US 1995-461395
                                                        19950605
                    A
                         19980714
    US 5746974
                    A 19980505
                                        US 1995-467028
                                                        19950606
    US 5798083
                     A 19980825
                                        US 1995-469464
                                                        19950606
                          20000620
                     Α
                                        US 1995-473313
                                                        19950607
    US 6078782
    US 2003008339
                          20030109
                                        US 2002-235127
                                                        20020905
                     A1
                                     US 1988-266882 A 19881103
PRIORITY APPLN. INFO.:
                                                    B2 19860430
                                     US 1986-858354
                                     US 1987-117017 B2 19871104
                                     EP 1989-913023 A3 19891031
                                                     A 19891031
                                     WO 1989-US4919
                                                     B2 19900618
                                     US 1990-539389
                                     US 1991-652427
                                                     B2 19910206
                                     US 1991-728093
                                                     B1 19910710
                                     US 1991-728194
                                                     B1 19910710
                                                     B1 19911115
                                     US 1991-792602
                                      US 1992-827269
                                                     B1 19920203
                                                     B1 19931124
                                     US 1993-158193
                                     US 1994-335183
                                                     B3 19941107
                                     US 1994-346832
                                                     A3 19941130
                                     US 1994-348749
                                                     B3 19941201
```

AB Homogeneous, electrochemiluminescent specific-binding assay methods are provided, as are reagent compns. and app. (diagrams included) for use in the methods. Binding of the electrochemiluminescent moiety-linked assay component to suspended microparticulate matter greatly modulates the intensity of the luminescent signal generated, providing a means of monitoring the specific binding reaction of the assay system. The suspended particles have little or no effect on the intensity of the luminescent signal generated by the electrochemiluminescent moiety-linked component which remains unbound to the suspended microparticulate matter. Thus, in the immunochem. detn. of theophylline, electrochemiluminescent signal counts were related to theophylline concn. in the range 0.0-40.0 .mu.g/mL. The methods of the invention were also used for immunochem. detn. of digoxin and IgGs, and in a hybridoma screening method.

L23 ANSWER 33 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN ACCESSION NUMBER: 1991:16852 HCAPLUS DOCUMENT NUMBER: 114:16852

TITLE: Use of time-dependent chemical sensor signals for

selective identification

AUTHOR(S): Maclay, G. Jordan; Stetter, Joseph R.; Christesen,

Dep. Electr. Eng. Comput. Sci., Univ. Illinois, CORPORATE SOURCE:

Chicago, IL, 60680, USA

Sensors and Actuators (1989), 20(3), 277-85 SOURCE:

CODEN: SEACDX; ISSN: 0250-6874

DOCUMENT TYPE: Journal LANGUAGE: English

A sensor system is described for identifying and quantifying different components in a gas mixt. by decoding information from a modulated output signal. The system operates by passing the unknown gas into a reaction chamber contg. a heated catalytic filament. The output products of the chamber are detected by an electrochem. sensor. The concn. of the reaction products is modulated by varying the temp. of the catalytic filament periodically. A theory is presented for analyzing the modulated output signal. The validity of the anal. is confirmed by expts. conducted with benzene, CO and HCN, and by computer simulations. It appears to be possible to identify different species in the inlet gas stream provided the species undergo chem. reactions in the reactor that have different activation energies. signal magnitude at peaks or valleys in the output is directly proportional to concn., even though a steady-state condition is not reached. The approach may be generalized to other detector systems.

L23 ANSWER 34 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1989:108542 HCAPLUS

DOCUMENT NUMBER: 110:108542

TITLE: Vasoconstrictor hormones depolarize renal glomerular

mesangial cells by activating chloride channels Kremer, Sid G.; Breuer, William V.; Skorecki, Karl L.

CORPORATE SOURCE: Dep. Med., Univ. Toronto, Toronto, ON, M5S 1A8, Can. Journal of Cellular Physiology (1989), 138(1), 97-105

CODEN: JCLLAX; ISSN: 0021-9541

DOCUMENT TYPE: Journal LANGUAGE: English

AUTHOR(S):

SOURCE:

The membrane potential-sensitive fluorescent dye bis-oxonol and the intracellular fluorescent Ca-sensitive probe Indo-1 were used to study the changes in membrane potential (Em) and intracellular free Ca concn. ([Ca2+]i) in cultured rat mesangial cells in response to vasoconstrictor hormones. Basal [Ca2+]i was 227 nM, and stimulation by maximal concns. of either vasopressin or angiotensin resulted in a transient 4-6-fold rise. Resting membrane potential was 45.8 mV and vasoconstrictor hormones caused a depolarization of 14-18 mV. The following extracellular ion substitutions indicated that Cl- efflux was the predominant ion flux responsible for depolarization: (1) depolarization persisted when Na in the medium was substituted with N-methylglucamine; (2) substitution of medium NaCl with Na gluconate, which enhances the gradient for Cl- efflux, augmented vasoconstrictor-stimulated depolarization; (3) suspension of cells in KCl medium resulted in depolarization, following which, stimulation by either vasopressin or angiotensin resulted in hyperpolarization; and (4) this hyperpolarization did not occur when K gluconate medium was used to depolarize the cells. The Ca ionophore ionomycin also resulted in membrane depolarization. However, prevention of the rise in [Ca2+]i by prior exposure to ionomycin in Ca-free medium or by loading mesangial cells with the intracellular Ca buffer BAPTA did not abrogate the depolarization response to vasoconstrictor hormones. Evidently, a rise in [Ca2+]i is not necessary for depolarization. In contrast, prior depolarization of the cells using varying concns. of KCl in the external medium, which dissipated the electrochem. gradient for Cl- efflux, resulted in a corresponding prolongation of the transient Ca response to vasopressin and angiotensin. Evidently,

angiotensin and vasopressin depolarize mesangial cells by activating C1channels and this activation can occur by both Ca-dependent and -independent mechanisms. In addn., activation of Cl- channels with resulting depolarization may serve to modulate the Ca signal.

L23 ANSWER (35\OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1988:403189 HCAPLUS

DOCUMENT NUMBER: 109:3189

TITLE: Fabrication and characterization of a

fiber-optic-based spectroelectrochemical probe

Van Dyke, David A.; Cheng, Hung Yuan AUTHOR(S):

Phys. Struct. Chem. Smith Kline and French Lab., CORPORATE SOURCE:

Philadelphia, PA, 19101, USA

Analytical Chemistry (1988), 60(13), 1256-60 CODEN: ANCHAM; ISSN: 0003-2700 SOURCE:

DOCUMENT TYPE: Journal LANGUAGE: English

Probes were constructed from fused silica optical fibers embedded in elec. conductive graphite/epoxy material with the optical and face and the working electrode active surface in a coplanar arrangement. electrochem. properties and spectroelectrochem. response of the microprobe were characterized in solns. and in gels contg. UV-absorbing oxidizable compds. The concept of electrochem. modulation of spectral signals for fiber-optic-assisted spectroscopy was demonstrated by using the ascorbate/dopamine gel model. The validity of such an approach for spectroelectrochem. measurements in tissue was evaluated by using isolated animal brain. Further development and possible applications are discussed.

L23 ANSWER 36 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1987:563887 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 107:163887

.Kinetics and mechanism of hydrogen and hydroxide TITLE:

electrosorption on faceted (100) platinum electrodes

in acid solutions

Bilmes, S. A.; Giordano, M. C.; Arvia, A. J. AUTHOR(S):

CORPORATE SOURCE: Inst. Invest. Fisicoquim. Teor. Apl., La Plata, 1900,

Argent.

Journal of Electroanalytical Chemistry and Interfacial SOURCE:

Electrochemistry (1987), 227(1-2), 183-98

CODEN: JEIEBC; ISSN: 0022-0728

DOCUMENT TYPE: Journal LANGUAGE: English

The multiplicity of voltammetric peaks related to H-adatom and OHelectroadsorption/electrodesorption reactions on faceted (100) Pt in 1M H2SO4 and 0.5M HClO4 were investigated at different temps. by using linear sweep voltammetry with triangular modulation covering a relatively large frequency range of the modulating signal. The voltammetric multiplicity related to H-adatom reactions in acid electrolytes is sensitive to both the surface structure and the structure and the structure of the soln. side of the elec. double layer. The kinetics of the reactions are apparently influenced in a cooperative way by the water-H-adatom-anion ensemble formed at each site of a particular crystallog. plane. The main 3 reactions exhibit different relaxation time consts. from which one can infer that the process occurring in the potential range presumably closer to the potential of zero charge is the slowest one. Anion adsorption produces a surface energy leveling effect and changes the kinetics of the electrochem. reactions. The H-adatom reactions can be interpreted through a formalism comprising the surface diffusion of H-adatom among different adsorption sites followed by a relatively fast electrochem. step involving only one kind of H-adatom. Data obtained for the OH--adsorbed reaction show, in principle,

the same trend as that of the H-adatom reactions, although in this respect more extensive data are required.

L23 ANSWER (37) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1986:567047 HCAPLUS

DOCUMENT NUMBER: 105:167047

TITLE: The sequence and energetics of cell membrane

transductive coupling to intracellular enzyme systems

AUTHOR(S): Adey, W. R.

Sch. Med., Loma Linda Univ., Loma Linda, CA, 92357, CORPORATE SOURCE:

USA

SOURCE: Bioelectrochemistry and Bioenergetics (1986), 15(3),

447-56

CODEN: BEBEBP; ISSN: 0302-4598

DOCUMENT TYPE: Journal; General Review

LANGUAGE:

English Weak oscillating electromagnetic fields in the pericellular environment AB

modulate key steps in coupling of signals from humoral stimuli at cell surface receptors to intracellular systems. Evidence that enzymic activity within the cell provides sensitive mol. markers about both the sequence and the energetics of transmembrane coupling mechanisms

is reviewed. A 3-stage model of membrane transductive coupling is proposed: a 1st stage in which weak pericellular electrochem.

oscillations and binding of humoral stimulating mols. at receptor sites initiates a highly cooperative modification of Ca binding, a 2nd stage involving transmission of signals initiated at receptor sites to the cell interior, and a 3rd stage dealing with intracellular response to the transmembrane signal. Exptl. data and models interrelating the pericellular electrochem. environment, cancer-promoting phorbol

esters, and activities of protein kinases and ornithine decarboxylase are discussed.

L23 ANSWER (38 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1986:431838 HCAPLUS

DOCUMENT NUMBER: 105:31838

TITLE: The application of modulated voltammetry in the study

of the corrosion and passivation of metals

Vilche, Jorge R.; Arvia, Alejandro J. AUTHOR(S):

CORPORATE SOURCE: Div. Electroquim., Inst. Invest. Fisicoquim. Tecr.

Apl., La Plata, Argent.

Anales de la Asociacion Quimica Argentina (1986), SOURCE:

74(2), 97-118

CODEN: AAQAAE; ISSN: 0365-0375

DOCUMENT TYPE: Journal LANGUAGE: English

Triangularly modulated both linear and triangular potential sweeps were

used to study the electrochem. behavior Cd, Co, Fe, Ni, and Ag in different aq. electrolytes at 25.degree.. From the corresponding complex E/I curves obtained through the systematic charge of the amplitude

and frequency of the modulating signal, qual.

evaluation of different intermediate species can be made. The

participation of chem. processes related to the stability of reactant and products related to film formation is envisaged. The advantages of this technique are discussed in comparison with other relaxation techniques.

L23 ANSWER (39 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1986:31010 HCAPLUS

DOCUMENT NUMBER: 104:31010

The lipid membrane as a selective chemical transducer TITLE:

Krull, Ulrich; Thompson, Michael AUTHOR(S):

Dep. Chem., Univ. Toronto, Toronto, ON, M5S 1A1, Can. CORPORATE SOURCE:

IEEE Transactions on Electron Devices (1985), SOURCE:

ED-32(7), 1180-4

CODEN: IETDAI; ISSN: 0018-9383

DOCUMENT TYPE: Journal; General Review

LANGUAGE: English

AB A review with 18 refs. about the use of bilayer lipid membranes as selective **electrochem**. based transducers for org. species and inorg. ions. The possible mechanisms which can be employed for signal generation are summarized. Operation hinges on a selective interaction between analyte in aq. soln. and membrane-embedded receptors. Complexation of these latter species results in phys. perturbation of the membrane structure and/or electrostatic fields, leading to the evolution of an anal. **signal** based on ion-current **modulation** through the membrane.

L23 ANSWER (40) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1983:146577 HCAPLUS

DOCUMENT NUMBER: 98:146577

TITLE: Mechanism of hole injection on ferric oxide

photoelectrodes AUTHOR(S): Pajkossy, Tamas

CORPORATE SOURCE: Cent. Res. Inst. Phys., Hung. Acad. Sci., Budapest,

H-1525, Hung.

SOURCE: Journal of the Electrochemical Society (1983), 130(3),

632-5

CODEN: JESOAN; ISSN: 0013-4651

DOCUMENT TYPE: Journal LANGUAGE: English

AB Photocurrent was measured on Fe2O3 photoelectrochem.—cell electrodes at const. electrode potential and slightly modulated illumination as a function of reducing agent concn. The av. value of the current was measured directly, and the current transients due to light modulation were recorded by using the signal averaging technique. The hole injection from the valence band is energetically possible. Due to insufficient data, it is not possible to give any est.

of the phenomenol. rate coeffs. as they depend on factors such as electrode potential, illumination intensity, nature of solute, etc.

L23 ANSWER 41 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1981:578000 HCAPLUS

DOCUMENT NUMBER: 95:178000

TITLE: Optical investigation of the electrical properties of

a polycrystalline-semiconductor-electrolyte interface

using electroreflectance

AUTHOR(S): Silberstein, R. P.; Lyden, Joseph K.; Tomkiewicz,

Micha; Pollak, Fred H.

CORPORATE SOURCE: Brooklyn Coll., City Univ. New York, Brooklyn, NY,

11210, USA

SOURCE: Journal of Vacuum Science and Technology (1981),

19(3), 406-10

CODEN: JVSTAL; ISSN: 0022-5355

DOCUMENT TYPE: Journal LANGUAGE: English

The low-field electrolyte electroreflectance (EER) spectra was studied for of polycryst. electrodeposited N-CdSe in the vicinity of the EO(A,B) transitions (direct gap at k.fwdarw. = 0). By using the properties of low-field EER, the distribution was studied of the applied a.c. voltage in situ in the photoelectrochem. solar cell configuration. By measuring the in-phase and quadrature EER signals as a function of modulation frequency f, the elec. impedance Z(f) was measured of the interface for 10 Hz < f <10 kHz, at the focus of the incident light. These results are in substantial agreement with direct elec. measurements. By measuring the EER amplitude as a function of applied d.c. bias voltage, evidence was found for surface states which contribute to Fermi level pinning; similar evidence was not obsd. for single-crystal

CdSe. These techniques are of general applicability for other semiconductor interfaces with optically transparent junctions, and are well suited to topog. scanning.

L23 ANSWER (42\OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1981:67666 HCAPLUS

DOCUMENT NUMBER: 94:67666

TITLE: Universal chronopotentiometer

Ashkinazi, V. E.; Drapkin, M. Ya.; Shestopalov, Yu. N. AUTHOR(S):

Vses. Nauchno-Issled. Konstr. Inst. Nauchn. CORPORATE SOURCE:

Priborostr., Leningrad, USSR

SOURCE: Gidrokhimicheskie Materialy (1979), 73, 99-103

CODEN: GKMAAS; ISSN: 0367-4665

DOCUMENT TYPE: Journal Russian LANGUAGE:

An exptl. model of a universal chronopotentiometer was developed. It

consists of a primary electrochem. converter, an elec. regulator, a signal modulator, an oscillog. recorder,

and a thermostat. Accuracy was .apprx.3% for detn. of  $0.05-0.5~\mathrm{mg}$  Pb/L water.

L23 ANSWER (43) OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1979:183393 HCAPLUS

90:183393 DOCUMENT NUMBER:

Anomalous oxygen uptake from isolated chloroplasts TITLE:

inhibited in photosystem II and without external

electron donors

AUTHOR(S): Muallem, Avner; Malkin, Shmuel

Biochem. Dep., Weizmann Inst. Sci., Rehovot, Israel CORPORATE SOURCE: SOURCE:

Biochimica et Biophysica Acta (1979), 546(1), 175-82

CODEN: BBACAQ; ISSN: 0006-3002

DOCUMENT TYPE: Journal LANGUAGE: English

A light-induced modulated signal of O uptake by isolated chloroplasts in the presence of Me viologen, when photosystem II activity was inhibited by DCMU and in the absence of external electron donors, was detected by a modulated O Pt electrode, polarized neg. An electrochem. process taking place on the surface of the neg. polarized Pt-cathode produces an intermediate which serves as an electron donor to Photosystem I. This intermediate may be the superoxide radical generated by the electrochem. redn. of O which continuously

L23 ANSWER 44 OF 44 HCAPLUS COPYRIGHT 2003 ACS on STN

1978:78688 HCAPLUS ACCESSION NUMBER:

88:78688 DOCUMENT NUMBER:

Electronic system for providing specificity in an TITLE:

diffuses from the external circulating medium to the electrode.

electrochemical analytical device David, Donald J.; Tucker, Huel C.

PATENT ASSIGNEE(S): United States Dept. of the Army, USA

U.S., 8 pp. SOURCE: CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

INVENTOR(S):

PATENT NO. KIND DATE APPLICATION NO. DATE -----\_\_\_\_\_\_ \_\_\_\_ US 4048041 · 19770913 US 1976-689287 19760524 Α US 1976-689287 19760524 PRIORITY APPLN. INFO.:

An electronic circuit is described which continuously provides specificity for an electrochem. cell for the monitoring of nitroglycerin (I)

[55-63-0] in wastewater by controlling the applied potential to the working electodes. The **electrochem**. cell has a Hg pool cathode, a Pt wire anode, and a calomel ref. electrode. A **modulated** neg. biased pulsed d.c. **signal** is used; the difference in the cell current at the start and finish of the pulse is measured with a demodulator with a differential amplifier circuit to give the current due only to the I redn. Current from any addnl. reactions that are const. or that occur above a desired redn. potential within the **electrochem**. cell are automatically cancelled by a gating circuit which is elec. coupled to the demodulator.